Access DB# 164845

# SEARCH REQUEST FORM

### Scientific and Technical Information Center

3	cientific and Technic	cai information Center								
Requester's Full Name:  Art Unit: 1763 Phone  Mail Box and Bldg/Room Location	10/2 GOUDY Number 300 172-1 on: Rex 7/22 Re	Examiner #: <u>69631</u> Date: <u>8-26-09</u> <u>43</u> 4 Serial Number: <u>10-807/39</u> sults Format Preferred (circle): PAPEN DISK E-MAI								
If more than one search is submitted, please prioritize searches in order of need.  **********************************										
Earliest Priority Filing Date:										
		t (parent, child, divisional, or issued patent numbers) along with the								
appropriate serial number.		· · · · · · · · · · · · · · · · · · ·								
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		Pat. & T.M. Office								
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STAFF USE ONLY	Type of Search	Vendors and cost where applicable								
Searcher: K, Fullin	NA Sequence (#)									
Searcher Phone #:	AA Sequence (#)	Dialog								
Searcher Location:	Structure (#)	Questel/Orbit								
Date Searcher Picked Up:	Bibliographic	Dr.Link								
Date Completed: 4/15/05	Litigation	Lexis/Nexis								
Searcher Prep & Review Time: 40	Fulltext	Sequence Systems								
Clerical Prep Time:	Patent Family	WWW/Internet								
Online Time: 7.7	Other	Other (specify)								

PTO-1590 (8-01)



# STIC Search Report

# STIC Database Tracking Number: 164045

TO: George A Goudreau

Location: 7A21 Art Unit: 1763

Socrab Notas

**September 16, 2005** 

Case Serial Number: 10/807139

From: Kathleen Fuller Location: EIC 1700 REMSEN 4B28

Phone: 571/272-2505

Kathleen.Fuller@uspto.gov

Search Notes	
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EIC17000

Questions about the scope or the results of the search? Contact the EIC searcher or contact:

Kathleen Fuller, EIC 1700 Team Leader 571/272-2505 REMSEN 4B28

Voluntary Results Feedback Form
<ul> <li>I am an examiner in Workgroup: Example: 1713</li> <li>Relevant prior art found, search results used as follows:</li> </ul>
☐ 102 rejection
103 rejection
Cited as being of interest.
Helped examiner better understand the invention.
Helped examiner better understand the state of the art in their technology.
Types of relevant prior art found:
Foreign Patent(s)
<ul> <li>Non-Patent Literature         (journal articles, conference proceedings, new product announcements etc.)     </li> </ul>
> Relevant prior art not found:
Results verified the lack of relevant prior art (helped determine patentability).
Results were not useful in determining patentability or understanding the invention.
Comments:

=> FILE HCAPLUS

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FILE COVERS 1907 - 15 Sep 2005 VOL 143 ISS 12 FILE LAST UPDATED: 14 Sep 2005 (20050914/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

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                OR 7440-21-3/BI OR 75-50-3/BI OR 75-59-2/BI OR 7631-86-9/BI)
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              2 SEA FILE=REGISTRY ABB=ON L2 AND PMS/CI
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L27
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L29
             9 SEA FILE=HCAPLUS ABB=ON L28 AND (CMP OR POLISH?)
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L32
            15 SEA FILE=HCAPLUS ABB=ON L31 AND (COMPOSITION? OR COMPNS)
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L33 \_\_21 SEA FILE=HCAPLUS ABB=ON L29 OR L32

=> FILE WPIX

FILE 'WPIX' ENTERED AT 17:14:12 ON 15 SEP 2005 COPYRIGHT (C) 2005 THE THOMSON CORPORATION

FILE LAST UPDATED: 15 SEP 2005 <20050915/UP>
MOST RECENT DERWENT UPDATE: 200559 <200559/DW>
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>>> FOR A COPY OF THE DERWENT WORLD PATENTS INDEX STN USER GUIDE, PLEASE VISIT:

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   FOR DETAILS. <<<</pre>

=> D QUE L34

L23

L24

L25

L26

L2 10 SEA FILE=REGISTRY ABB=ON (106392-12-5/BI OR 12033-89-5/BI OR 121-44-8/BI OR 1310-58-3/BI OR 1336-21-6/BI OR 691397-13-4/BI OR 7440-21-3/BI OR 75-50-3/BI OR 75-59-2/BI OR 7631-86-9/BI) L3 1 SEA FILE=REGISTRY ABB=ON SILICA/CN T.4 2 SEA FILE=REGISTRY ABB=ON L2 AND PMS/CI L6 709369 SEA FILE=HCAPLUS ABB=ON L3 OR SILICA OR SIO2 OR SILICON ?OXIDE L11 2454 SEA FILE=HCAPLUS ABB=ON L6 AND NONION? (2A) SURFACT? L1244 SEA FILE=HCAPLUS ABB=ON L11 AND (CMP OR CHEM? (1A) MECH? (1A) POLI SH?) L1321 SEA FILE=HCAPLUS ABB=ON L12 AND SLURR? L15 65429 SEA FILE=HCAPLUS ABB=ON L4 OR ((PROPYLENE OXIDE OR PO) AND (EO OR ETHYLENE OXIDE)) OR NONION? (2A) SURFACT? L16 375 SEA FILE=HCAPLUS ABB=ON L15 AND ABRASIV? 54 SEA FILE=HCAPLUS ABB=ON L16 AND SLURR? 1.17 24 SEA FILE=HCAPLUS ABB=ON L17 AND (CMP OR CHEM? (1A) MECH? (1A) POLI L18 SH?) L19 25030 SEA FILE=HCAPLUS ABB=ON L4 OR ((PROPYLENE OXIDE OR PO) AND (EO OR ETHYLENE OXIDE)) L20 1607 SEA FILE=HCAPLUS ABB=ON L6 AND L19 L21 14 SEA FILE=HCAPLUS ABB=ON L20 AND (CMP OR CHEM? (1A) MECH? (1A) POLI SH?) L22 39 SEA FILE=HCAPLUS ABB=ON L20 AND SLURR?

7 SEA FILE=HCAPLUS ABB=ON L22 AND POLISH?

4 SEA FILE=REGISTRY ABB=ON L2 AND 1-2/N

49123 SEA FILE=HCAPLUS ABB=ON L25

45 SEA FILE=HCAPLUS ABB=ON L13 OR L18 OR L21 OR L23

L35

1 SEA FILE=COMPENDEX ABB=ON L29 OR L32

=> FILE JICST

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                121-44-8/BI OR 1310-58-3/BI OR 1336-21-6/BI OR 691397-13-4/BI
                OR 7440-21-3/BI OR 75-50-3/BI OR 75-59-2/BI OR 7631-86-9/BI)
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L4
              2 SEA FILE=REGISTRY ABB=ON L2 AND PMS/CI
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L6
                ?OXIDE
           2454 SEA FILE=HCAPLUS ABB=ON L6 AND NONION? (2A) SURFACT?
L11
L12
             44 SEA FILE=HCAPLUS ABB=ON L11 AND (CMP OR CHEM? (1A) MECH? (1A) POLI
L13
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L15
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L16
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L32
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L36
              O SEA FILE=JICST-EPLUS ABB=ON L29 OR L32
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### => FILE INSPEC

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=> D QUE L37
             10 SEA FILE=REGISTRY ABB=ON (106392-12-5/BI OR 12033-89-5/BI OR
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                121-44-8/BI OR 1310-58-3/BI OR 1336-21-6/BI OR 691397-13-4/BI
                OR 7440-21-3/BI OR 75-50-3/BI OR 75-59-2/BI OR 7631-86-9/BI)
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              1 SEA FILE=REGISTRY ABB=ON SILICA/CN
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L19
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           1607 SEA FILE=HCAPLUS ABB=ON L6 AND L19
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             49 SEA FILE=HCAPLUS ABB=ON L30 AND (POLISH? OR CMP)
L32
            15 SEA FILE=HCAPLUS ABB=ON L31 AND (COMPOSITION? OR COMPNS)
1.37
            1 SEA FILE=INSPEC ABB=ON L29 OR L32
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## => DUP REM L33 L34 L35 L37

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PROCESSING COMPLETED FOR L35
PROCESSING COMPLETED FOR L37

Polymers, uses

IT

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34 DUP REM L33 L34 L35 L37 (3 DUPLICATES REMOVED)
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=> D L39 ALL 1-34 HITSTR L39 ANSWER 1 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN DUPLICATE 1 AN 2005:527317 HCAPLUS applicant DN 143:69836 ED Entered STN: 19 Jun 2005 ΤI Slurry compositions and CMP methods using IN Choi, Jaekwang; Lee, Jaedong; Hong, Chang-Ki S. Korea PA SO U.S. Pat. Appl. Publ., 17 pp. CODEN: USXXCO DT Patent LΑ English IC ICM H01L021-76 ICS H01L021-302; H01L021-461 INCL 438690000 76-3 (Electric Phenomena) Section cross-reference(s): 48, 66 FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE . ---------**----**ΡI US 2005130428 20050616 A1 US-,2004-807139 20040324 JP 2005175498 A2 20050630 JP 2004-359039 20041210 PRAI KR 2003-90551 Α 20031212 US 2004-807139 20040324 Α CLASS CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. ---------US 2005130428 ICM H01L021-76 ICS H01L021-302; H01L021-461 INCL 438690000 US 2005130428 NCL 438/690.000 JP 2005175498 FTERM 3C058/AA07; 3C058/CB01; 3C058/CB10; 3C058/DA02; 3C058/DA12; 3C058/DA17 AB The exemplary embodiments of the present invention providing new slurry compns. suitable for use in processes involving the chemical mech. polishing (CMP) of a polysilicon layer. The slurry compns. include one or more nonionic polymeric surfactants that will selectively form a passivation layer on an exposed polysilicon surface to suppress the polysilicon removal rate relative to Si oxide and Si nitride and improve the planarity of the polished substrate. Exemplary surfactants include alkyl and aryl alcs. of ethylene oxide (EO) and propylene oxide ( PO) block copolymers and may be present in the slurry compns. in an amount of up to .apprx.5%, although much smaller concns. may be effective. Other slurry additives may include viscosity modifiers, pH modifiers, dispersion agents, chelating agents, and amine or imine surfactants suitable for modifying the relative removal rates of Si nitride and Si oxide. ST surfactant block copolymer CMP polysilicon semiconductor device fabrication

RL: TEM (Technical or engineered material use); USES (Uses)

(block; slurry compns. and CMP methods to

Ethylene oxide-propylene oxide block

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GOUDREAU 10/807139 09/16/2005
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Page 8

copolymer 106392-12-5D, Ethylene oxidepropylene oxide block copolymer, monononylphenyl and
monoctylphenyl ethers 691397-13-4, Ethylene
oxide-propylene oxide triblock copolymer
RL: TEM (Technical or engineered material use); USES (Uses)
(slurry compns. and CMP methods to
suppress polysilicon removal)
75-50-3 HCAPLUS
Methanamine, N,N-dimethyl- (9CI) (CA INDEX NAME)

RN

CN

RN 75-59-2 HCAPLUS CN Methanaminium, N,N,N-trimethyl-, hydroxide (9CI) (CA INDEX NAME)

● OH -

RN 121-44-8 HCAPLUS CN Ethanamine, N,N-diethyl- (9CI) (CA INDEX NAME)

RN 1336-21-6 HCAPLUS CN Ammonium hydroxide ((NH4)(OH)) (9CI) (CA INDEX NAME)

 $H_4N-OH$ 

RN 106392-12-5 HCAPLUS CN Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)

CM 1

CRN 75-56-9 CMF C3 H6 O

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CH<sub>3</sub>
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CM 2

CRN 75-21-8 CMF C2 H4 O



RN 106392-12-5 HCAPLUS

CN Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)

CM 1

CRN 75-56-9 CMF C3 H6 O

Сн3

CM 2

CRN 75-21-8 CMF C2 H4 O



RN 691397-13-4 HCAPLUS CN Oxirane, methyl-, po

Oxirane, methyl-, polymer with oxirane, triblock (9CI) (CA INDEX NAME)

CM 1

CRN 75-56-9

CMF C3 H6 O



CM 2

CRN 75-21-8

CMF C2 H4 O



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ANSWER 2 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
     2005:823155 HCAPLUS
AN
ED
     Entered STN: 19 Aug 2005
ΤI
     Synergistic antibacterial formulation containing cefixime trihydrate,
     cloxacillin sodium and Lactobacillus sporogenes spores
IN
     Khandelwal, Sanjeev
PA
SO
     U.S. Pat. Appl. Publ., 25 pp.
     CODEN: USXXCO
DT
     Patent
     English
LA
     ICM A61K031-545
IC
     ICS A61K031-4λ; A61K031-43; A61K031-716
INCL 424470000; 5141\2000; 514057000; 514200000
     63-6 (Pharmaceut\icals)
     Section cross-reference(s): 1
FAN.CNT 1
     PATENT NO.
                        KIND
                              DATE
                                          APPLICATION NO.
                                                                  DATE
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                         - - - -
                               -----
                                           -----
PI
     US 2005181051
                         A1
                               20050818
                                           US 2004-13110
                                                                  20041215
                               20050824 EP 2005-250879
     EP 1566176
                         A1
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK,
            BA, HR, IS, YU
PRAI IN 2004-MU178
                               20040216
     IN 2004-MU258
                               20040303
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
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US 20050181051 ICM
                       A61K031\545
                ICS
                       A61K031-47; A61K031-43; A61K031-716
                INCL
                       424470000; 514192000; 514057000; 514200000
US 2005181051
                NCL
                       424/470.0\0; 514/192.000; 514/057.000; 514/200.000
    A synergistic antibacterial formulation for oral delivery of cefixime
AB
     trihydrate, cloxacillin sodium in an extended release form and an
     immediate release form, and Lactobacillus sporogenes spores is provided.
     For example, sustained-release granules were prepared by wet granulation of
     cloxacillin sodium 50.0 kg and hydroxypropyl Me cellulose (HPMC; average
    viscosity 4000 cps) 6.0 kg , usilg a binder comprising HPMC (average viscosity
     50 cps) 800g dissolved in a mixture of dichloromethane 8.0 kg and iso-Pr
     alc. 12.0 kg. The core was prepared by blending cloxacillin sodium
     sustained-release granules obtained with a mixture of cloxacillin sodium
     particle 7.6 kg, cefixime trihydrate\particles 11.2 kg, L. sporogenes
     spores 750 g, sodium starch glycollate 1.0 kg, colloidal silicon
     dioxide 0.3 kg, sodium lauryl sulfate \1.0 kg and talc 1.0 kg was
     prepared Magnesium stearate 1.0 kg was added and further blendded,
     resulting in the lubricated core mass. This core mass was then compressed
     into cores of average weight of 806.2 mg/plus/minus>3%. The core obtained were
    pan coated with a film coating composition containing Et cellulose 0.8
     kg, hydroxypropyl cellulose 0.8 kg, iso-Pr\alc. 12 kg, methylene chloride
     22 kg, di-Et phthalate 0.01 kg and titanium dioxide 0.15 kg in a stainless
     steel container and stirred for five minute's using overhead stirrer until
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ST

IT

Cation exchangers Chondrus crispus

Combination chemotherapy

a smooth slurry was obtained. The coated tablets were polished with talc. The film-coated tablet (average weight 820 mg <plus/minus>3%) contained (i) cloxacillin sodium equivalent to 250 mg cloxacillin sustained release, (ii) cloxacillin sodium equivalent to 250 mg cloxacillin immediate release, (III) cefixime trihydrate equivalent to 100 mg cefixime immediate release, and (IV) L. sporogenes 45 million spores. cloxacillin cefixime Lactobacillus coated tablet synergistic antibacterial Fatty acids RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (C6-30; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Alcohols RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (C16-18, ethoxylated; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Alcohols Glycerides RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (C16-18; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Gums and Mucilages (Panwar gum, larch arabolactan; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Drug delivery systems (controlled-release; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Monoglycerides RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (diacetylated; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Alcohols RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (fatty, C6-30; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Lipids RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (glycerolipids; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Citrus (pulp; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) (silica; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes Waxes RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (spermaceti; synergistic antibacterial formulation containing cefixime trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores) Acacia Antibacterial agents Antibiotics Bacillus coagulans Beeswax

Stearyl Alcohol

446-72-0, Genistein 480-40-0, Chrysin

```
Human
    Molasses
    Porifera
     Surfactants
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
IT
    Alcohols
    Carnauba wax
    Ceramides
     Cottonseed oil
     Flavonoids
    Gelatins
     Glycerides
     Polyamides
     Polyoxyalkylenes
     Polysiloxanes
     Rosin
       Silica qel
     Smectite-group minerals
     Sovbean oil
     Sphingolipids
     Sphingosines
     Waxes
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
IT
    Drug interactions
        (synergistic; synergistic antibacterial formulation containing cefixime
        trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores)
IT
    Drug delivery systems
        (tablets, coated; synergistic antibacterial formulation containing cefixime
        trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores)
IT
     9003-39-8D, crosslinked
    RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (Crospovidone; synergistic antibacterial formulation containing cefixime
        trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores)
IT
    7631-86-9, Colloidal silicon dioxide
    RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (colloidal; synergistic antibacterial formulation containing cefixime
        trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores)
IT
     642-78-4, Cloxacillin Sodium
                                  125110-14-7, Cefixime Trihydrate
    RL: ADV (Adverse effect, including toxicity); PAC (Pharmacological
     activity); PKT (Pharmacokinetics); THU (Therapeutic use); BIOL (Biological
     study); USES (Uses)
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
IT
     9004-65-3, Hydroxypropyl methylcellulose
    RL: PRP (Properties); THU (Therapeutic use); BIOL (Biological study); USES
     (Uses)
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
TΤ
     50-70-4D, Sorbitol, derivs.
                                  50-99-7, Glucose
                                                      56-81-5D, Glycerol,
              57-10-3, Palmitic Acid
                                       57-11-4, Stearic acid
                                                                57-50-1,
    Sucrose
              57-50-1D, Sucrose, esters 57-88-5, Cholesterol
                                                                  63-42-3,
               67-68-5, Dimethyl sulfoxide 69-65-8D, Mannitol, derivs.
    71-41-0, Pentanol 106-11-6, Diethylene glycol monostearate
     1,2-Ethanediol, derivs.
                               110-54-3, Hexane 111-03-5, Glyceryl monooleate
    111-27-3, Hexanol 111-60-4, Ethylene glycol monostearate 112-92-5,
```

117-39-5, Quercetin 151-21-3, Sodium lauryl sulfate

480-44-4, Acacetin

IT

RN

CN

IT

RN

CN

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538-24-9D, Glyceryl trilaurate, glyceryl ester
                                                                546-93-0,
     Magnesium carbonate 555-45-3D, Glyceryl trimyristate, glyceryl ester
                                    557-05-1, Zinc stearate
     557-04-0, Magnesium stearate
                                                              1309-48-4,
                      1323-39-3, Propylene glycol monostearate
     Magnesium oxide
                                                                  1338-39-2,
                          1338-41-6, Sorbitan monostearate
     Sorbitan monolaurate
                                                               1338-43-8,
     Sorbitan monooleate
                           1344-95-2, Calcium silicate
                                                         1592-23-0, Calcium
                3097-08-3, Magnesium lauryl sulfate
                                                      7757-93-9, Dibasic
                        7758-87-4, Tribasic calcium phosphate
     calcium phosphate
                                                                8007-43-0,
     Sorbitan sesquioleate
                             9000-01-5, Gum arabic
                                                    9000-28-6, Ghatti gum
     9000-30-0, Guar gum
                           9002-18-0, Agar
                                             9002-89-5, Polyvinyl alcohol
     9002-92-0, Lauromacrogol
                                9003-20-7, Polyvinyl acetate
                                                               9003-39-8,
     Polyvinylpyrrolidone 9004-32-4, Carboxymethylcellulose sodium
     9004-34-6, Cellulose
                            9004-38-0, Cellulose acetate phthalate
     Dextrin
               9004-57-3, Ethylcellulose
                                         9004-64-2, Hydroxypropylcellulose
     9004-67-5, Methylcellulose
                                  9004-99-3, PEG Stearate
                                                           9005-25-8, Starch
     9005-38-3, Sodium alginate
                                  9005-64-5, polysorbate 20
                                                              9005-65-6,
     polysorbate 80
                     9005-66-7, polysorbate 40
                                                 9005-67-8, polysorbate 60
     9005-70-3, polysorbate 85
                                 9005-71-4, polysorbate 65
                                                            9011-14-7,
     Polymethyl methacrylate
                              9036-19-5
                                          9036-66-2, Arabinogalactan
               9057-02-7, Pullulan 9057-06-1, Carboxymethyl starch
     9050-04-8
     9063-38-1, Sodium starch glycolate
                                        10236-47-2, Naringin
                                                                11099-07-3D,
     Glyceryl stearate, glyceryl ester
                                        12619-70-4, Cyclodextrin
                                                                    14807-96-6,
     Talc
            14987-04-3, Magnesium trisilicate
                                               25087-26-7, Polymethacrylic
     acid
            25301-02-4, Tyloxapol
                                    25322-68-3, Polyethylene glycol
     25322-68-3D, Macrogol, ethers and esters 25618-55-7D, Polyglycerol,
               26027-38-3, Nonoxynol
                                       26266-57-9, Sorbitan monopalmitate
     26266-58-0, Sorbitan trioleate
                                     26338-57-8, Maleic anhydride-vinyl
     alcohol copolymer
                        26658-19-5, Sorbitan tristearate
     Glyceryl monostearate
                           37353-59-6, Hydroxymethyl cellulose
                                                                   54182-62-6,
     Polacrilin
                 58858-21-2, Hydroxypropyl methylcellulose acetate
     69670-80-0, Hydroxymethyl propylcellulose
                                               74811-65-7, Croscarmellose
              86157-80-4 106392-12-5, Poloxamer
                                                  222414-51-9,
     Polyethylene glycol hydroxystearate
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
     7631-86-9, Colloidal silicon dioxide
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (colloidal; synergistic antibacterial formulation containing cefixime
        trihydrate, cloxacillin sodium and Lactobacillus sporogenes spores)
     7631-86-9 HCAPLUS
     Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)
o = si = o
     106392-12-5, Poloxamer
     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
        (synergistic antibacterial formulation containing cefixime trihydrate,
        cloxacillin sodium and Lactobacillus sporogenes spores)
     106392-12-5 HCAPLUS
     Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)
     CM
          1
    CRN 75-56-9
     CMF C3 H6 O
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CH<sub>3</sub>
```

CM 2

CRN 75-21-8 CMF C2 H4 O



```
ANSWER 3 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     2005-416664 [42]
AN
                        WPIX
DNN
    N2005-338024
                        DNC C2005-127515
     Slurry useful for chemical mechanical
     polishing in e.g. manufacturing semiconductor device comprises a
     copper oxidizing agent; a complexing agent; a surfactant; an inorganic
     particle; and resin particle containing polystyrene.
DC
     A97 E19 G04 L02 L03 P61 U11
     FUKUSHIMA, D; MINAMIHABA, G; YAMAMOTO, S; YANO, H
IN
     (TOKE) TOSHIBA KK; (FUKU-I) FUKUSHIMA D; (MINA-I) MINAMIHABA G; (YAMA-I)
PA
     YAMAMOTO S; (YANO-I) YANO H
CYC
PΙ
     US 2005118821
                     A1 20050602 (200542) *
                                                14
                                                      H01L021-302
     JP 2005159166
                   A 20050616 (200542)
                                                19
                                                      H01L021-304
ADT
    US 2005118821 A1 US 2004-932096 20040902; JP 2005159166 A JP 2003-398163
     20031127
PRAI JP 2003-398163
                          20031127
IC
     ICM H01L021-302; H01L021-304
     ICS B24B037-00; C09K003-14; H01L021-461
AB
     US2005118821 A UPAB: 20050704
    NOVELTY - A chemical mechanical polishing (
     CMP) slurry comprises a copper oxidizing agent; a
     complexing agent forming a copper organic complex; a surfactant; an
     inorganic particle; and a resin particle (R1) containing polystyrene. (R1)
    Has a functional group of same kind of polarity as that of the inorganic
    particle, an average particle diameter of less than 100 nm, and is
     incorporated at a concentration of less than 1 weight%.
         DETAILED DESCRIPTION - INDEPENDENT CLAIMS are included for the
     following:
          (1) polishing of a semiconductor involving contacting a
    polishing surface of the semiconductor substrate with a
    polishing pad attached to a turntable, and dropping the
    CMP slurry on the polishing pad to
    polish the polishing surface; and
          (2) manufacturing a semiconductor device involving forming an
```

insulating film above a semiconductor substrate, forming a recessed portion in the insulating film, depositing a conductive material inside the recessed portion and above the insulating film to form a conductive layer, and removing the conductive material which is deposited above the

selectively leave the conductive material in the recessed portion.

(CMP) of a copper film formed on a surface of a semiconductor

insulating film by CMP using a CMP slurry to

USE - For chemical mechanical polishing

FS

FA MC

AN

ΤI

DC

IN PΑ

CYC PΙ

ADT

IC

AB

09/16/2005 Page 15 substrate having an over-plating portion, by removing the residual copper in the region of over-plating to form a copper film of uniform thickness, in the manufacture of a semiconductor device. ADVANTAGE - By using the CMP slurry composition the polishing can be performed at high polishing rate while suppressing the generation of erosion or dishing. Dwq.0/6 CPI EPI GMPI AB; DCN CPI: A04-C02E; A12-A03; E06-D01; E06-D02; E06-D08; E07-D04C; E10-A09B4; E10-B02D6; E10-C02D1; E10-C02D2; E10-E04M3; E31-E01; E31-E03; E31-P03; E34-C02; E35-K02; G04-B04; L02-F; L02-J; L04-C27 EPI: U11-A10; U11-C06A1A ANSWER 4 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN 2005-402873 [41] WPIX DNN N2005-326740 DNC C2005-124404 Manufacture of semiconductor device, e.g. memory, comprises chemical mechanical polishing spin-on glass film on silicon dioxide film using slurry containing cerium oxide and cationic surfactant. MATSUI, Y; MINAMIHABA, G; SHIGETA, A; YANO, H (TOKE) TOSHIBA KK; (MATS-I) MATSUI Y; (MINA-I) MINAMIHABA G; (SHIG-I) SHIGETA A; (YANO-I) YANO H US 2005106874 A1 20050519 (200541) \* 26 H01L021-302 JP 2005109452 A 20050421 (200541) 21 H01L021-304 KR 2005027157 A 20050317 (200557) H01L021-304 US 2005106874 A1 US 2004-935429 20040908; JP 2005109452 A JP 2004-258030 20040906; KR 2005027157 A KR 2004-72936 20040913 PRAI JP 2004-258030 20040906; JP 2003-321474 20030912 ICM H01L021-302; H01L021-304 ICS H01L021-461 US2005106874 A UPAB: 20050629 NOVELTY - Manufacturing a semiconductor device comprises depositing a silicon dioxide film (3) on a substrate (1) having a wiring pattern (2); coating a spin-on glass (SOG) film (4) on the silicon dioxide film; and polishing the spin-on glass film using slurry containing cerium oxide and cationic surfactant with a chemical-mechanical polishing process. DETAILED DESCRIPTION - A further INDEPENDENT CLAIM is included for manufacture of a semiconductor device. USE - For manufacturing a semiconductor device e.g. memory, high speed logic large scale integrated circuit (LSI), system LSI or memory/logic mixed LSI. ADVANTAGE - The planarization step executed with the chemical -mechanical polishing process that uses the

slurry containing cerium oxide and cationic surfactant(s) selectively polishes SiO2 film and SOG film with the polishing selectivity value of at least 5. The reliability of the semiconductor device can be enhanced through the enhancement of the insulating capability of the semiconductor device.

DESCRIPTION OF DRAWING(S) - The figure shows a section of a semiconductor device during processing. Substrate 1

Wiring pattern 2

Silicon dioxide film 3

```
SOG film 4
     Dwg.3B/16
     CPI EPI
FS
FA
     AB; GI
MC
     CPI: L04-B04A; L04-C12; L04-C12D; L04-C27; L04-E15
     EPI: U11-C06A1A; U11-C18B5
     ANSWER 5 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
L39
AN
     2005-354123 [36]
                        WPIX
DNN
     N2005-289162
                        DNC C2005-109369
ΤI
     Slurry useful for chemical mechanical
     polishing of copper and barrier films comprises at least one
     abrasive, at least one organic phosphonate, at least one oxidizer
     and water.
DC
     A97 E11 L02 L03 U11
IN
     JENG, W; TSENG, S; YANG, K
PA
     (JENG-I) JENG W; (TSEN-I) TSENG S; (YANG-I) YANG K
CYC
PΙ
     US 2005090104
                    A1 20050428 (200536)*
                                                12
                                                      H01L021-302
ADT US 2005090104 Al Provisional US 2003-514830P 20031027, US 2004-958417
     20041006
PRAI US 2003-514830P
                          20031027; US 2004-958417
                                                          20041006
IC
     ICM H01L021-302
     ICS H01L021-461
AB
     US2005090104 A UPAB: 20050608
     NOVELTY - A chemical mechanical polishing
     slurry comprises at least one abrasive, at least one
     organic phosphonate, at least one oxidizer and water.
          USE - For chemical mechanical polishing
     of copper and barrier films (claimed).
          ADVANTAGE - The slurry is capable of polishing
     copper at high removal rate at a relatively low down force so that
     CMP throughput is improved and CMP defects are reduced,
     has high selectivity to tantalum barrier. The barrier slurry
     delivers good planarity, has high hydrogen peroxide stability so that
     slurry pot lifetime is extended and the performance of
     chemical mechanical polishing process is
     stable. The slurry gives low dishing and erosion so that the
     interconnect metal lines have uniform sheet resistance and the wafer
     surface planarity is good for fabrication multi-level interconnects. The
     slurry provides smooth copper surface on the polished
     surface, gives low copper corrosion defects on polished wafers,
     is cost effective. The organic phosphonates are excellent chelating agents
     for copper. The phosphonates generally have significantly higher
     solubility in water comparing to many carboxylic and amino acids or salts
     given in the prior art for copper chemical mechanical
     polishing. The manufacturing process of slurry is
     usually simpler and more cost effective.
     Dwg.0/5
     CPI EPI
FS
     AB; DCN
FA
MC
     CPI: A12-W12B; E05-G02; E05-G03D; E31-C; E31-E01; E31-E03; E31-P03;
          E34-C02; E34-E; E35-K02; E35-S; E35-U05; L02-F04; L04-B04A; L04-C27
     EPI: U11-A10; U11-C06A1A
L39
    ANSWER 6 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
ΑN
     2005-151610 [16]
                        WPIX
CR
     2005-519376 [53]
DNN
    N2005-127934
                        DNC C2005-048924
ΤI
     Chemical mechanical polishing slurry
```

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for removing silicon dioxide in preference to silicon
     nitride useful in manufacture of semiconductor wafer, comprises
     selectivity enhancer comprising nucleic acid related compound e.g.
     uridine.
     A85 B04 D16 P61 U11
DC
IN
     HEGDE, S; XU, W; XU, W P
     (HEGD-I) HEGDE S; (XUWW-I) XU W; (NYAC-N) NYACOL NANO TECHNOLOGIES INC
PA
CYC
     108
PΙ
     US 2005028450
                     A1 20050210 (200516) *
                                                       B24D003-02
     WO 2005014746
                     A1 20050217 (200516)
                                           EN
                                                       C09G001-02
        RW: AT BE BG BW CH CY CZ DE DK EA EE ES FI FR GB GH GM GR HU IE IT KE
            LS LU MC MW MZ NA NL OA PL PT RO SD SE SI SK SL SZ TR TZ UG ZM ZW
         W: AE AG AL AM AT AU AZ BA BB BG BR BW BY BZ CA CH CN CO CR CU CZ DE
            DK DM DZ EC EE EG ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG
            KP KR KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NA NI NO NZ
            OM PG PH PL PT RO RU SC SD SE SG SK SL SY TJ TM TN TR TT TZ UA UG
            US UZ VC VN YU ZA ZM ZW
ADT
     US 2005028450 A1 US 2003-635949 20030807; WO 2005014746 A1 WO 2004-US25536
     20040806
PRAI US 2003-635949
                          20030807
     ICM B24D003-02; C09G001-02
IC
     ICS C09K003-14; H01L021-304; H01L021-306
AΒ
     US2005028450 A UPAB: 20050818
     NOVELTY - A chemical mechanical polishing (
     CMP) slurry comprises a liquid, abrasive
     particles and a selectivity enhancer comprising a nucleic acid related
     compound.
          DETAILED DESCRIPTION - An INDEPENDENT CLAIM is also included for a
     CMP polishing process for selectively removing
     silicon dioxide form the surface of workpiece containing
     surface areas of silicon dioxide and surface areas of
     silicon nitride in which the surface to be polished is contacted
     with a polishing pad and a CMP slurry is
     applied to the interface between the polishing pad the surface
     to be polished.
          USE - The slurry is used for selectively removing
     silicon dioxide from the surface of a work-piece
     containing surface areas of silicon dioxide and
     surface areas of silicon nitride. It is useful in the manufacture of
     semiconductor wafers and chips.
          ADVANTAGE - The slurry exhibits high selectivities for
     removing silicon dioxide in preference to silicon
     nitride.
     Dwg.0/0
     CPI EPI GMPI
FS
FA
     AB; DCN
MC
     CPI: A12-A03; A12-E07C; B04-B03A; B04-B03B; B04-B03D; B04-B03E; B04-C03;
          B04-E01; B05-B02C; B06-D09; B07-D04A; B07-D12; B10-A22; B10-B04B;
          D05-H10; D05-H13
     EPI: U11-A10; U11-C06A1A
L39
    ANSWER 7 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     2005-120520 [13]
ΑN
                        WPIX
CR
     2004-641155 [62]
DNN
    N2005-103959
                        DNC C2005-040088
ΤI
     Chemical mechanical polishing slurry
     used in forming damascene wiring used for manufacturing semiconductor
     device, comprises solvent, abrasive grains, and silicone-based
     surfactant having specific hydrophilic lipophilic balance value.
DC
     A97 G04 L03 U11
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INCL

438689000

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IN
     MATSUI, Y; MINAMIHABA, G; YANO, H
PA
     (TOKE) TOSHIBA KK
CYC
   1
    US 2005009322 A1 20050113 (200513)* 15 H01L021-4763
PΙ
ADT US 2005009322 Al Div ex US 2003-706052 20031113, US 2004-909287 20040803
FDT US 2005009322 A1 Div ex US 6794285
PRAI JP 2003-37179
                        20030214
     ICM H01L021-4763
     ICS H01L021-302; H01L021-461
AB
    US2005009322 A UPAB: 20050224
    NOVELTY - A chemical mechanical polishing (
     CMP) slurry comprises a solvent, abrasive
     grains, and silicone-based surfactant having an hydrophilic lipophilic
     balance (HLB) value of 7-20.
         USE - The CMP slurry is used in the formation of
     damascene wiring used for mounting a high-speed logic large scale
     integrated circuit (LSI), system LSI and memory/logic hybrid LSI. It is
     used in the manufacture of semiconductor device.
         ADVANTAGE - The slurry allows the formation of damascene
     wiring, where the density of defects and the concentration of surface
     impurities can be minimized.
         DESCRIPTION OF DRAWING(S) - The figure shows a cross-sectional view
     of a method of manufacturing a semiconductor device.
     Plug 102
         Laminated insulating films 103, 104
         Barrier metal film 105
         Wiring material film 106
    Dwg.6A/6
    CPI EPI
FS
    AB; GI
FA
MC
    CPI: A05-H01B; A06-A00E; A12-A03; A12-W12C; G04-B04; L04-C13B; L04-C27
    EPI: U11-A10; U11-C05D2; U11-C06A1A
L39
    ANSWER 8 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
    2004:392243 HCAPLUS
DN
    140:398393
ED
    Entered STN: 14 May 2004
TI
    Chemical mechanical polishing
    composition and method
IN
    Li, Yuzhou; Keleher, Jason; Zhao, Junzi; Brancewicz, Chris
PA
    Sachem, Inc., USA
SO
    U.S. Pat. Appl. Publ., 31 pp.
    CODEN: USXXCO
DT
    Patent
LΑ
    English
ΙÇ
    ICM C09K003-14
    ICS H01L021-302; H01L021-461
INCL 438689000
CC
    76-3 (Electric Phenomena)
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                       APPLICATION NO.
    -----
                       ----
                                         -----
PI
    US 2004092102
                       A1
                              20040513
                                         US 2002-292404
                                                               20021112
PRAI US 2002-292404
                              20021112 /
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 -----
               _____
US 2004092102
               ICM
                      C09K003-14
                ICS
                     H01L021-302; H01L021-461
```

US 2004092102 NCL 438/689.000 **ECLA** C09G001/04; H01L021/321P2 AB A chemical-mech. polishing (CMP) formulation and method for using the same. The composition is useful for polishing semiconductor substrates, and particularly substrate surfaces containing copper, tungsten, or alloys of the same. CMP formulation may contain a copolymer enhancement agent such as a Pluronics compound (e.g., Pluronic P103, Pluronic P123, Pluronic F108, Pluronic F88, etc.), and/or a vesicle encapsulating agent, as well as an active agent that is chemical reactive with the substrate to enhance polishing performance. The active agent may be a bifunctional compound that is capable of functioning as both a passivating agent and a complexing agent to achieve an optimum rate of passivation and oxidation on the substrate surface. An active agent can also take the form of an oxidation activator, such as a metal ion, encapsulated in a vesicle or micelle, that is released with applied pressure to accelerate the removal process and improve planarization efficiency. ST chem mech polishing compn process IT Abrasives Diodes Hydrotropes Integrated circuits Interconnections, electric Oxidizing agents Transistors (chemical-mech. polishing composition and polishing method using same) IT Amines, uses Phospholipids, uses RL: NUU (Other use, unclassified); USES (Uses) (chemical-mech. polishing composition and **polishing** method using same) IT Polishing (chemical-mech.; chemical-mech. polishing composition and polishing method using same) IT Solvents (cosolvents; chemical-mech. polishing composition and polishing method using same) IT Acids, uses RL: NUU (Other use, unclassified); USES (Uses) (inorg.; chemical-mech. polishing composition and polishing method using same) IT Surfactants (nonionic; chemical-mech. polishing composition and polishing method using same) IT Acids, uses RL: NUU (Other use, unclassified); USES (Uses) (organic; chemical-mech. polishing composition and polishing method using same) IT Sulfonic acids, uses RL: NUU (Other use, unclassified); USES (Uses) (salts, aryl; chemical-mech. polishing composition and polishing method using same) IT Semiconductor materials (substrate; chemical-mech. polishing composition and polishing method using same) IT Copper alloy, base Tungsten alloy, base

RL: DEV (Device component use); PEP (Physical, engineering or chemical

```
process); PYP (Physical process); PROC (Process); USES (Uses)
        (chemical-mech. polishing composition
       and polishing method using same)
IT
     1306-38-3, Ceria, uses
                            1309-48-4, Magnesia, uses 1314-23-4, Zirconia,
           1344-28-1, Alumina, uses
    RL: NUU (Other use, unclassified); USES (Uses)
        (abrasive; chemical-mech. polishing
       composition and polishing method using same)
                                     7440-25-7, Tantalum, processes
IT
     7429-90-5, Aluminum, processes
     7440-32-6, Titanium, processes 7440-33-7, Tungsten, processes
     7440-50-8, Copper, processes
     RL: DEV (Device component use); PEP (Physical, engineering or chemical
    process); PYP (Physical process); PROC (Process); USES (Uses)
        (chemical-mech. polishing composition
       and polishing method using same)
IT
     51-17-2, Benzimidazole
                            56-40-6, Glycine, uses 64-18-6, Formic acid,
           64-19-7, Acetic acid, uses 67-63-0, 2-Propanol, uses
     Propanoic acid, uses
                           107-21-1, Ethylene glycol, uses 112-60-7,
    Tetraethylene glycol
                          144-62-7, Oxalic acid, uses 147-85-3, Proline,
           288-32-4D, Imidazole, derivative 1300-72-7, Sodium xylenesulfonate
     1310-73-2, Sodium hydroxide, uses 7631-86-9, Silicon
     oxide, uses 7647-01-0, Hydrochloric acid, uses 7664-41-7,
    Ammonia, uses 7664-93-9, Sulfuric acid, uses 7697-37-2, Nitric acid,
          7732-18-5, Water, uses 10034-85-2, Hydrogen iodide
     106392-12-5, Pluronic P103
                                203945-07-7, SiLK (dielectric)
    RL: NUU (Other use, unclassified); USES (Uses)
        (chemical-mech. polishing composition
       and polishing method using same)
IT
     7722-84-1, Hydrogen peroxide, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (chemical-mech. polishing composition
       and polishing method using same)
IT
     7631-86-9, Silicon oxide, uses
     106392-12-5, Pluronic P103
    RL: NUU (Other use, unclassified); USES (Uses)
        (chemical-mech. polishing composition
       and polishing method using same)
     7631-86-9 HCAPLUS
RN
CN
     Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)
o = si = o
    106392-12-5 HCAPLUS
RN
CN
    Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)
    CM
         1
    CRN
         75-56-9
     CMF
         C3 H6 O
```



CM 2

CRN 75-21-8 CMF C2 H4 O



IT

Surfactants

```
ANSWER 9 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
    2004:759257 HCAPLUS
DN
    141:252819
ED
    Entered STN: 17 Sep 2004
    Aluminum or glass substrates for magnetic hard disks, their manufacture,
ΤI
    and polishing slurries
IN
    Horie, Yuji; Okuyama, Hiromitsu; Tanifuji, Tatsuya
    Nippon Micro Coating Co., Ltd., Japan
PA
SO
    Jpn. Kokai Tokkyo Koho, 16 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IC
    ICM G11B005-73
    ICS B24B021-00; B24B037-00; C09K003-14; G11B005-84
CC
    77-8 (Magnetic Phenomena)
    Section cross-reference(s): 57
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                        APPLICATION NO.
                                                                DATE
                              -----
     -----
                       ----
                                          -----
PΙ
    JP 2004259417
                        A2
                              20040916
                                          JP 2003-92680
                                                                20030224
    US 2004241379
                        A1
                              20041202
                                         US 2004-776372
PRAI JP 2003-92680
                       Α
                              20030224
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 ------
                ____
                      -----
 JP 2004259417
                ICM
                       G11B005-73
                       B24B021-00; B24B037-00; C09K003-14; G11B005-84
                ICS
JP 2004259417
                FTERM 3C058/AA05; 3C058/AA07; 3C058/AA09; 3C058/AC04;
                       3C058/CB01; 3C058/CB03; 3C058/DA17; 5D006/CB04;
                       5D006/CB07; 5D006/DA03; 5D006/EA04; 5D112/AA02;
                       5D112/AA24; 5D112/BA06; 5D112/GA09; 5D112/GA13;
                       5D112/GA14
US 2004241379
                NCL
                       428/848.200
                ECLA
                      G11B005/73N; G11B005/84B
AΒ
    The substrates are equipped with textured streaks of 70 lines/µm in the
    radius direction. Method for manufacture of the substrates includes
    polishing the substrates by pressing a running polishing
    tape towards the substrates that are rotating in the opposite direction,
    under feeding a polishing slurry. The
    polishing slurry contains single crystal diamond
    particles, polycryst. diamond particles, or their clusters having diameter
    1-50 nm as abrasive grains and water or water-based solution as
    dispersants. The said polishing slurries containing
    clusters of the single crystal and/or polycryst. diamond particles are
    also claimed.
ST
    aluminum substrate polishing magnetic hard disk; glass substrate
    streaked surface hard disk; diamond cluster particle abrasive
    polishing slurry
```

(anionic, slurry dispersant; manufacture of Al (alloy) or glass

substrates with grooved surfaces for magnetic hard disks, by polishing with diamond cluster abrasive slurries)

IT Clusters

(diamond particle; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster **abrasive slurries**)

IT Amides, uses

RL: TEM (Technical or engineered material use); USES (Uses) (fatty, slurry dispersant; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by polishing with diamond cluster abrasive slurries)

IT Surface structure

(grooved; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster **abrasive slurries**)

IT Magnetic disks

(hard; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster **abrasive slurries**)

IT Glass substrates

Polishing

(manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster **abrasive slurries**)

IT Fatty acids, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(metal salts, slurry dispersant; manufacture of Al (alloy) or
glass substrates with grooved surfaces for magnetic hard disks, by
polishing with diamond cluster abrasive
slurries)

IT Surfactants

(nonionic, slurry dispersant; manufacture of Al (alloy)
or glass substrates with grooved surfaces for magnetic hard disks, by
polishing with diamond cluster abrasive
slurries)

IT Polyamide fibers, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(polishing cloth; manufacture of Al (alloy) or glass substrates
with grooved surfaces for magnetic hard disks, by polishing
with diamond cluster abrasive slurries)

IT Slurries

(polishing; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by polishing with diamond cluster abrasive slurries)

IT Glycols, uses

Phosphates, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(slurry dispersant; manufacture of Al (alloy) or glass substrates
with grooved surfaces for magnetic hard disks, by polishing
with diamond cluster abrasive slurries)

IT Amines, uses

RL: TEM (Technical or engineered material use); USES (Uses) (vegetable oil, slurry dispersant; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by polishing with diamond cluster abrasive slurries)

IT Fats and Glyceridic oils, uses

RL: TEM (Technical or engineered material use); USES (Uses)

(vegetable, amines, slurry dispersant; manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by polishing with diamond cluster abrasive slurries)

IT. Aluminum alloy, base

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster **abrasive slurries**)

IT 7782-40-3, Diamond, uses

RL: TEM (Technical or engineered material use); USES (Uses)
(abrasive particles; manufacture of Al (alloy) or glass substrates
with grooved surfaces for magnetic hard disks, by polishing
with diamond cluster abrasive slurries)

IT 7429-90-5, Aluminum, processes

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(manufacture of Al (alloy) or glass substrates with grooved surfaces for magnetic hard disks, by **polishing** with diamond cluster abrasive slurries)

- L39 ANSWER 10 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
- AN 2004:198610 HCAPLUS
- DN 140:244785
- ED Entered STN: 11 Mar 2004
- TI Chemical-mechanical polishing process for forming wiring structures, and abrasive compositions used therein
- IN Sakai, Kenji; Tamai, Kazumasa; Kawamura, Atsunori; Matsuda, Takeshi; Hirano, Tatsuhiko; Ina, Katsuyoshi
- PA Fujimi, Inc., Japan
- SO Jpn. Kokai Tokkyo Koho, 22 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese
- IC ICM C09K003-14

ICS B24B037-00; H01L021-304

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 37, 57

FAN.CNT 1

	PATENT NO.			KIND	DATE		API	PLICATION	NO.	DATE
		<del></del>						· <b></b>		
ΡI	JP	20040758	62	A2	200403	11	JP	2002-2385	96	20020819
	US	20040844	14	A1	200405	06	US	2003-6429	129 <sup>.</sup>	20030818
PRAI	JP	2002-238	596	Α	200208	19				
CLASS	3									
PATI	ENT	NO.	CLASS	PATENT	FAMILY	CLASSI	FIC	CATION COL	ES	

JP 2004075862 ICM C09K003-14

71 2004075002 ICM CO5R005 14

ICS B24B037-00; H01L021-304

JP 2004075862 FTERM 3C058/AA07; 3C058/CB01; 3C058/DA02; 3C058/DA13;

3C058/DA17 US 2004084414 NCL 216/088.000

ECLA C09G001/02; C23F003/00

OS MARPAT 140:244785

AB The process for **polishing** of a material having a Cu-based elec. conductive layer formed on a barrier layer on an elec. insulating layer

ST

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having concave parts, involves (1) polishing of the elec.
conductive layer not to expose the barrier layer with an abrasive
composition containing abrasives (A) selected from SiO2 and
Al203, polishing accelerators (B) selected from glycine and
\alpha-alanine, H2O, and H2O2, (2) polishing of the elec.
conductive layer to expose the barrier layer with an abrasive
composition containing the abrasives (A), the polishing
accelerators (B), organic compds. (C) selected from poly(ethylene
oxide) , poly(propylene oxide) , polyoxyethylene
alkyl ethers, polyoxypropylene alkyl ethers, polyoxyethylene-
polyoxypropylene alkyl ethers, and polyoxyalkylene addition polymers having
carbon triple bonds R1O(X)mCR3R5C.tplbond.CCR4R6(Y)nOR2 (R1-R6 = H, C1-10
alkyl; X, Y = ethyleneoxy, propyleneoxy; m, n = 1-20), corrosion
inhibitors (D) selected from benzotriazole and its derivs., H2O2, and H2O,
and (3) polishing of the barrier layer with an abrasive
composition containing the abrasives (A), acids (E) selected from HNO3,
HCl, lactic acid, H3PO4, H2SO4, AcOH, oxalic acid, citric acid, tartaric
acid, malonic acid, succinic acid, maleic acid, and fumaric acid or
alkalies (F) selected from KOH, NH4OH, and NaOH, the corrosion inhibitors
(D), and H2O. The process prevents dishing or erosion.
chem mech polishing abrasive compn
wiring; dishing erosion prevention chem mech
polishing
Alcohols, uses
RL: NUU (Other use, unclassified); TEM (Technical or engineered material
use); USES (Uses)
   (alkoxylated; chemical-mech. polishing
   process and abrasive compns. for forming wiring structures
   without dishing or erosion)
Polyoxyalkylenes, uses
RL: NUU (Other use, unclassified); TEM (Technical or engineered material
use); USES (Uses)
   (alkyl ethers; chemical-mech. polishing
   process and abrasive compns. for forming wiring structures
   without dishing or erosion)
Abrasives
Corrosion inhibitors
Electric conductors
  Polishing materials
Semiconductor device fabrication
   (chemical-mech. polishing process and
   abrasive compns. for forming wiring structures without
   dishing or erosion)
Acids, uses
Bases, uses
Polyoxyalkylenes, uses
Polyoxyalkylenes, uses
RL: NUU (Other use, unclassified); TEM (Technical or engineered material
use); USES (Uses)
   (chemical-mech. polishing process and
   abrasive compns. for forming wiring structures without
   dishing or erosion)
Polishing
   (chemical-mech.; chemical-mech.
   polishing process and abrasive compns. for forming
   wiring structures without dishing or erosion)
Alcohols, uses
RL: NUU (Other use, unclassified); TEM (Technical or engineered material
use); USES (Uses)
```

(ethoxylated; chemical-mech. polishing

IT

ΙT

IT

IT

ΙT

09/16/2005 Page 25 process and abrasive compns. for forming wiring structures without dishing or erosion) Polyoxyalkylenes, uses RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses) (mono (alkyl group) - terminated; chemical-mech. polishing process and abrasive compns. for forming wiring structures without dishing or erosion) 7440-50-8, Copper, processes RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process); USES (Uses) (chemical-mech. polishing process and abrasive compns. for forming wiring structures without dishing or erosion) 50-21-5, Lactic acid, uses 56-40-6, Glycine, uses  $\alpha$ -Alanine, uses 64-19-7, Acetic acid, uses 77-92-9, Citric acid, 87-69-4, Tartaric acid, uses 110-15-6, Succinic acid, uses 110-16-7, Maleic acid, uses 110-17-8, Fumaric acid, uses 141-82-2, Malonic acid, uses 144-62-7, Oxalic acid, uses 1310-58-3, Potassium hydroxide, uses 1310-73-2, Sodium hydroxide, uses 1336-21-6, Ammonium hydroxide 1344-28-1, Aluminum oxide, uses 7647-01-0, Hydrochloric acid, uses 7664-38-2, Phosphoric acid, uses 7664-93-9, Sulfuric acid, 7697-37-2, Nitric acid, uses 7722-84-1, Hydrogen peroxide, uses 9003-11-6D, Ethylene oxide-propylene oxide copolymer, alkyl ethers 9014-85-1 25322-68-3, Poly( ethylene oxide) 25322-68-3D, Polyethylene glycol, alkyl ethers 25322-69-4, Poly(propylene oxide) 25322-69-4D, Polypropylene glycol, alkyl ethers RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses) (chemical-mech. polishing process and abrasive compns. for forming wiring structures without dishing or erosion) 7631-86-9, Colloidal silica, uses RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses) (colloidal; chemical-mech. polishing process and abrasive compns. for forming wiring structures without dishing or erosion) 95-14-7, 1H-Benzotriazole RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses) (corrosion inhibitor; chemical-mech. polishing process and abrasive compns. for forming wiring structures without dishing or erosion) 7631-86-9, Colloidal silica, uses (colloidal; chemical-mech. polishing process

IT

RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses)

and abrasive compns. for forming wiring structures without

dishing or erosion)

7631-86-9 HCAPLUS RN

CN Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

o = si = o

L39 ANSWER 11 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN

magnetic heads)

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AN
     2004:159198 HCAPLUS
     140:191477
DN
    Entered STN: 27 Feb 2004
ED
    Lapping oil compositions for finish polishing of
TI
     composites constituted by multiple different-hardness materials,
     especially thin-film magnetic heads
IN
    Saito, Isao; Orii, Kazuya
PA
    Tokyo Magnetic Printing Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 9 pp.
     CODEN: JKXXAF
DT
     Patent
LA
    Japanese
     ICM B24B037-00
IC
     ICS C09K003-14; G11B005-31
     77-8 (Magnetic Phenomena)
CC
     Section cross-reference(s): 38, 57
FAN.CNT 1
     PATENT NO.
                        KIND
                              DATE
                                          APPLICATION NO.
                                                                  DATE
     -----
                        ----
                                           -----
    JP 2004058220
                         A2
                               20040226
                                           JP 2002-220920
                                                                  20020730
PRAI JP 2002-220920
                               20020730
CLASS
PATENT NO.
                CLASS PATENT FAMILY CLASSIFICATION CODES
                ICM
JP 2004058220
                       B24B037-00
                ICS
                       C09K003-14; G11B005-31
JP 2004058220 FTERM 3C058/AA07; 3C058/AC04; 3C058/CA01; 3C058/CB02;
                       3C058/DA02; 5D033/DA22; 5D033/DA31
os
    MARPAT 140:191477
AB
    The lapping oil compns., free from abrasive grains,
    contain nonag, solvents and amine additives. The composites, e.g.,
    metal-ceramic composites, subjected for main polishing with
    dropping nonag.-solvent slurries containing nonionic
     surfactants as disperse medium and free abrasive grains,
    are finish polished with the lapping oil compns. with
    dropping the compns. The lapping oil compns. can
    easily exclude unfixed abrasive grains for achieving high
    precise finish polishing.
    finish polishing lapping oil additive amine; composite finish
ST
    polishing lapping oil additive amine; magnetic head finish
    polishing lapping oil additive amine; metal ceramic composite
    finish polishing lapping oil
IT
    Alcohols, uses
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
        (amino; lapping oil compns. containing amine additives
       for finish polishing of composites, especially thin-film magnetic
       heads)
IT
    Metals, processes
    RL: PEP (Physical, engineering or chemical process); PYP (Physical
    process); PROC (Process)
        (composites containing; lapping oil compns. containing amine
       additives for finish polishing of composites, especially thin-film
       magnetic heads)
IT
    Amines, uses
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
        (ethoxylated; lapping oil compns. containing amine
       additives for finish polishing of composites, especially thin-film
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KR 2004050726

EP 1570512

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ΙT
     Polishing
        (finish; lapping oil compns. containing amine additives
        for finish polishing of composites, especially thin-film magnetic
        heads)
IT
     Ceramic composites
     Magnetic recording heads
       Polishing materials
        (lapping oil compns. containing amine additives for
        finish polishing of composites, especially thin-film magnetic
        heads)
     Amines, uses
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (secondary; lapping oil compns. containing amine
        additives for finish polishing of composites, especially thin-film
        magnetic heads)
     Amines, uses
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (tertiary; lapping oil compns. containing amine
        additives for finish polishing of composites, especially thin-film
        magnetic heads)
IT
     51427-90-8, Iron alloy, Fe, Al, Si
     RL: DEV (Device component use); PEP (Physical, engineering or chemical
     process); PYP (Physical process); PROC (Process); USES (Uses)
        (Sendust, in magnetic heads; lapping oil compns. containing
        amine additives for finish polishing of composites,
        especially thin-film magnetic heads)
IT
                  113724-99-5, Aluminum titanium carbide oxide
     RL: DEV (Device component use); PEP (Physical, engineering or chemical
     process); PYP (Physical process); PROC (Process); USES (Uses)
        (in magnetic heats; lapping oil compns. containing amine
        additives for finish polishing of composites, especially thin-film
        magnetic heads)
IT
     110-58-7, Amylamine
                           111-86-4, Octylamine
                                                  112-18-5,
     Dodecyldimethylamine
                            112-90-3, Oleylamine
                                                   124-22-1, Dodecylamine
     124-30-1, Stearylamine
                              52811-24-2
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
        (lapping oil compns. containing amine additives for
        finish polishing of composites, especially thin-film magnetic
       heads)
    ANSWER 12 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
L39
    2004-487944 [46]
AN
                        WPIX
DNN
    N2004-384927
                        DNC C2004-181839
TΙ
    Slurry composition for secondary polishing
    of silicon wafers, comprises hydroxyalkylcellulose-based water-soluble
    polymeric thickener and polyoxyethylenealkylamine ether-based
    nonionic surfactant.
DC
    A11 A25 A85 A97 E19 G04 L03 U11
IN
    LEE, G S; LEE, I G; NOH, H S; PARK, T W; LEE, I K; LEE, K S; ROH, H S
PA
     (CHEI-N) CHEIL IND INC
CYC
    31
ΡI
    WO 2004053968
                     A1 20040624 (200446) * EN
                                                14
                                                      H01L021-304
       RW: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LU MC NL PT RO
            SE SI SK TR
         W: CN JP US
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EN

H01L021-304

H01L021-304

A 20040617 (200466)

A1 20050907 (200559)

```
R: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI LU MC NL PT
            RO SE SI SK TR
     WO 2004053968 A1 WO 2003-KR1532 20030730; KR 2004050726 A KR 2002-77860
ADT
     20021209; EP 1570512 A1 EP 2003-812702 20030730, WO 2003-KR1532 20030730
     EP 1570512 A1 Based on WO 2004053968
FDT
PRAI KR 2002-77860
                          20021209
IC
     ICM H01L021-304
AB
     WO2004053968 A UPAB: 20040720
     NOVELTY - A slurry composition comprises (weight%)
     colloidal silica having an average particle diameter of 30-80 nm
     as an abrasive (2-10), ammonia (0.5-1.5), a
     hydroxyalkylcellulose-based water-soluble polymeric thickener (0.2-1), a
     polyoxyethylenealkylamine ether-based nonionic
     surfactant (0.03-0.5), a quaternary ammonium base (0.01-1), and
     deionized water (balance).
          USE - For secondary polishing of silicon wafers during
     chemical mechanical polishing (CMP)
     process.
          ADVANTAGE - The composition is capable of increasing the
     dispersion stability of silica as an abrasive to
     improve the polishing quality on pitted microscratches and
     reducing the amount of silica particles to lower the
     manufacturing cost. It has a low concentration of silica and
     high dispersibility.
     Dwg.0/0
FS
     CPI EPI
     AB; DCN
FA
MC
     CPI: A03-A04A; A12-E07C; E07-D05; E10-A22E; E10-A22G; E10-B03B3; E31-P03;
          E32-A02; G04-B08; L04-C27
     EPI: U11-A10; U11-C06A1A
L39
    ANSWER 13 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP ON STN
AN
     2004-784203 [77]
                        WPIX
DNN
    N2004-618064
                        DNC C2004-274426
     Aqueous solution for use in processing semiconductor wafers; comprises
TI
     block copolymer surfactant and abrasive particles of silica,
     alumina, and/or ceria.
DC
     A25 A26 A85 L03 U11
IN
     COOPER, K E; FLAKE, J C; GROSCHOPF, J; SOLOMENTSEV, Y E
PA
     (COOP-I) COOPER K E; (FLAK-I) FLAKE J C; (GROS-I) GROSCHOPF J; (SOLO-I)
     SOLOMENTSEV Y E
CYC
    1
PΙ
     US 2004224426
                     A1 20041111 (200477) *
                                                 7
                                                      H01L021-00
ADT
    US 2004224426 A1 US 2003-430987 20030507
PRAI US 2003-430987
                          20030507
IC
     ICM H01L021-00
     ICS C09K003-14
AB
     US2004224426 A UPAB: 20041203
     NOVELTY - Aqueous solution comprises a block copolymer surfactant having a
     hydrophobic portion and a hydrophilic portion, and abrasive particles. The
     abrasive particles comprise silica, alumina, and/or ceria.
          DETAILED DESCRIPTION - An INDEPENDENT CLAIM is also included for a
     method for processing semiconductor wafers comprising providing a
     semiconductor wafer (150) having a wafer surface to a semiconductor
     processing stage, and applying a block copolymer surfactant having a
     hydrophobic portion and a hydrophilic portion to the wafer surface during
     the semiconductor processing stage.
          USE - For use in processing semiconductor wafers (claimed).
          ADVANTAGE - The inventive aqueous solution is a reliable solution
     that minimizes surface defects and does not require large additive
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concentrations. The surfactant in the aqueous solution is a surface-active agent that reduces surface tension of a liquid or surface tension between a liquid and a solid. DESCRIPTION OF DRAWING(S) - The figure illustrates a schematic of a chemical mechanical polishing processing tool that can be used in processing semiconductor wafers. Chemical mechanical polishing tool 100 Semiconductor wafer 150 Polish film 155 Drive assembly 191 Actuator assembly 192 Dwg.1/2 FS CPI EPI AB; GI FΑ MC CPI: A12-H10; A12-W12C; L04-C07C; L04-C07F; L04-C09; L04-C27 EPI: U11-A10; U11-C06A1A L39 ANSWER 14 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN 2004-286736 [27] WPIX AN DNN N2004-227376 DNC C2004-110441 Application composition for manufacture of insulating thin film, TIcontains organic polymer, organic solvent(s) and silica precursor containing compound(s) chosen from alkoxysilane and its hydrolysis substance, and polycondensate. DC A26 A85 L03 U11 PA (ASAH) ASAHI KASEI KK CYC PΙ JP 2004018608 A 20040122 (200427)\* 28 C09D183-04 ADT JP 2004018608 A JP 2002-173542 20020614 PRAI JP 2002-173542 20020614 IC ICM C09D183-04 ICS C09D001-00; C09D005-25; C09D171-02; C09D183-02; H01L021-316; H01L021-768 AB JP2004018608 A UPAB: 20040426 NOVELTY - Application composition contains silica precursor, organic polymer and organic solvent(s) chosen from alcohol, ketone, amide and ester. The silica precursor contains compound(s) chosen from 1-6 functional alkoxysilane and its hydrolysis substance, and polycondensate. The organic polymer contains aliphatic ether block copolymer of binary or ternary. DETAILED DESCRIPTION - Application composition contains silica precursor, organic polymer and at least one type of organic solvent selected from alcohol, ketone, amide and ester. The silica precursor contains at least one type of compound chosen from 1-6 functional alkoxysilane and its hydrolysis substance, and a polycondensate. The alkoxysilane is of formula (1) and/or formula (2): R1n(Si)(OR2)4-nR3m(R40)3-mSi-(R7)p-Si(OR5)3-qR6q(2) R1-R6 = hydrogen or univalent organic group; = 0-3;m and q = 0-2; R7 = group with oxygen atom or (CH2)r; = 1-6; and = 0 or 1.The sum of silicon atom originating in the alkoxysilane of 1-3 functionality and its hydrolysis substance, and polycondensate with respect to the total of the silicon atom originating in the alkoxysilane and its hydrolysis substance, is 5-80 mol%. The organic polymer contains

10 weight% or more of aliphatic ether block copolymer of binary or ternary

except poly(ethylene oxide)-poly(propylene

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AB

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oxide) -poly(ethylene oxide) which is a ternary
    block copolymer, with respect to all organic polymers. The block copolymer
    is of formula: ((R80)x-(R100)y-(R90)z).
         R8-R10 = 1-10C alkylene group;
      = 2-200;
    y = 2-100; and
    z = 0-200.
         INDEPENDENT CLAIMS are included for the following:
          (1) insulating thin film;
          (2) wiring structure using the insulating thin film as insulator; and
          (3) semiconductor element with wiring structure.
         USE - For manufacture of insulating thin film for wiring structure
    used for semiconductor element (claimed) and large scale integrated
    multilayer interconnections.
         ADVANTAGE - The porous thin film with stable and low dielectric
    constant, is formed using the application composition. The thin
    film has mechanical strength which endures chemical
    mechanical polishing process in copper-wiring process of
    semiconductor element, and hence suitable for insulating films of
    substrate for large scale integration multilayer interconnections or
    semiconductor element.
    Dwg.0/0
    CPI EPI
    AΒ
    CPI: A05-H01B; A06-A00E2; A08-S02; A12-E07C; L04-C12E
    EPI: U11-C05A
    ANSWER 15 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
    2003:48\1849 HCAPLUS
    139:40498
    Entered STN: 24 Jun 2003
    Abrasive compositions for CMP of device wafers
    Miyazaki, Tadakazu
    Sanyo Chemical Industries, Ltd., Japan
    Jpn. Kokai Tokkyo Koho, 8 pp.
    CODEN: JKXXAF
    Patent
    Japanese
    ICM C09K003-14
    ICS B24B037-00; H01L021-304
    57-7 (Ceramics)
    Section cross-reference(s): 76
FAN.CNT 1
    PATENT NO.
                        KIND
                              DATE
                                        APPLICATION NO.
                                                                DATE
    -----
                        _ _ _ _
                              -----
                                          -----
                                                                -----
    JP 2003176479
                        A2
                              20030624
                                          JP 2002-280113
                                                               20020926
PRAI JP 2001-298001
                        Α
                              20010927
CLASS
PATENT NO.
               CLASS PATENT FAMILY CLASSIFICATION CODES
               ----
 -----
JP 2003176479
               ICM
                      C09K003-14
                ICS
                      B24B037-00; H01L021-304
    The abrasive compns. contain polyoxyalkylene ethers represented
    by general formula Z[[(CH2CH2O)a(AO)b]R]p [R = C1-18 alkenyl, C2-18
    alkenyl or alkapolyenyl, C2-24 acyl, H; Z = H-removed residues of C1-12
    p-valent alcs.; A = C3-4 alkylene; a, b = 1-100 integer; units of
    (CH2CH2O) and (AO) are randomly bonded or block bonded; weight ratio of
    CH2CH2O group and AO group = 30/70-70/30' p = 1-6 integer], aliphatic
    carboxylic acid amides, and water. The abrasives may comprise inorg.
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compds. selected from SiO2, Al2O3, Ce oxide, Si3N4, and ZrO2.

In the process, device wafers with wirings being formed are polished with the abrasive compns. Scratches on wafer surfaces have been suppressed.

chem mech polishing abrasive compn surfactant; semiconductor device wafer CMP abrasive compn; polyoxyalkylene nonionic surfactant chem mech polishing; aliph carboxylic acid amide surfactant CMP

IT Abrasives

ST

(abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT Amides, uses

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(aliphatic; abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT Polishing

(chemical-mech.; abrasive compns. containing
polyoxyalkylene ethers and aliphatic carboxylic acid amides for
CMP of device wafers)

IT Polyoxyalkylenes, uses

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(ethers; abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT Surfactants

(nonionic; abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT 93-83-4, Oleic acid diethanolamide 111-58-0, Oleic acid monoethanolamide 106392-12-5, Ethylene oxide-propylene oxide block copolymen

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT 1306-38-3, Cerium oxide, uses 1314-23-4, Zirconium oxide, uses 1344-28-1, Alumina, uses 12033-89-5, Silicon nitride, uses RL: NUU (Other use, unclassified); USES (Uses)

(abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT 7631-86-9, Colloidal silica, uses

RL: NUU (Other use, unclassified); USES (Uses)

(colloidal; abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

IT 106392-12-5, Ethylene oxide-propylene
 oxide block copolymer

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(abrasive compns. containing polyoxyalkylene ethers and aliphatic carboxylic acid amides for CMP of device wafers)

RN 106392-12-5 HCAPLUS

CN Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)

CM 1

CRN 75-56-9 CMF C3 H6 O

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O CH<sub>3</sub>
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CM 2

CRN 75-21-8 CMF C2 H4 O

 $\angle$ 

RN 7631-86-9 HCAPLUS CN Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

o = si = o

L39 ANSWER 16 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN

AN 2003:443964 HCAPLUS

DN 139:15944

ED Entered STN: 10 Jun 2003

TI Coating composition for manufacture of electrically insulating porous silica film used in wiring structure of semiconductor device

IN Hanahata, Hiroyuki

PA Asahi Kasei Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 15 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09D183-02

ICS C01B033-12; C09D001-00; C09D005-25; C09D153-00; C09D171-02; C09D183-04; H01L021-312; H01L021-316

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 38

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE ------\_\_\_\_ ----**---**----------JP 2003165952 PΙ A2 20030610 JP 2001-364582 20011129 PRAI JP 2001-364582 20011129

CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

TD 000000000

JP 2003165952 ICM C09D183-02

ICS C01B033-12; C09D001-00; C09D005-25; C09D153-00; C09D171-02; C09D183-04; H01L021-312; H01L021-316

OS MARPAT 139:15944

AB The composition comprises (1) silica precursors containing (A)

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Si(OR1)4 (R1 = divalent organic group), its hydrolyzate, and/or its
polycondensate and/or (B) R2Si(OR1)3 (R1, R2 = divalent organic group), its
hydrolyzate, and/or its polycondensate to satisfy mol. fraction of
(B-derived Si)/[(A-derived Si) + (B-derived Si)] 1-50 mol% (excluding 50
mol%), (2) organic polymers containing linear or branched block copolymers, and
(3) solvents of alcs., ketones, amides, and/or esters. The porous
SiO2 film is manufactured by coating a substrate with the compn
., converting the silica precursors to a gel to give a
silica-organic polymer composite film, and removing the organic polymers
from the composite film. The porous film has low dielec. constant and high
durability in chemical mech. polishing in Cu
wiring process, and scarcely generates pollutant gases in via-hole
silica precursor org polymer porous film insulator; alkoxysilane
block copolymer porous silica film manuf
Silanes
RL: RCT (Reactant); RACT (Reactant or reagent)
   (alkoxy; alkoxysilane-organic block copolymer compns. for manufacture
   of elec. insulating porous silica film in semiconductor
   device)
Dielectric films
Semiconductor devices
Sol-gel processing
   (alkoxysilane-organic block copolymer compns. for manufacture of
   elec. insulating porous silica film in semiconductor device)
Porous materials
   (films; alkoxysilane-organic block copolymer compns. for manufacture
   of elec. insulating porous silica film in semiconductor
   device)
Films
   (porous; alkoxysilane-organic block copolymer compns. for manufacture
   of elec. insulating porous silica film in semiconductor
   device)
7631-86-9P, Silica, uses
RL: DEV (Device component use); IMF (Industrial manufacture); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
   (alkoxysilane-organic block copolymer compns. for manufacture of
   elec. insulating porous silica film in semiconductor device)
132894-01-0, Ethylene oxide-propylene
oxide block copolymer dimethyl ether
RL: NUU (Other use, unclassified); REM (Removal or disposal); PROC
(Process); USES (Uses)
   (alkoxysilane-organic block copolymer compns. for manufacture of
   elec. insulating porous silica film in semiconductor device)
78-10-4, Tetraethoxysilane
                             2031-67-6, Methyltriethoxysilane
RL: RCT (Reactant); RACT (Reactant or reagent)
   (alkoxysilane-organic block copolymer compns. for manufacture of
   elec. insulating porous silica film in semiconductor device)
24991-55-7, Polyethylene glycol dimethyl ether
RL: NUU (Other use, unclassified); REM (Removal or disposal); PROC
(Process); USES (Uses)
   (block copolymers mixed with; alkoxysilane-organic block copolymer
   compns. for manufacture of elec. insulating porous silica
   film in semiconductor device)
7631-86-9P, Silica, uses
RL: DEV (Device component use); IMF (Industrial manufacture); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
   (alkoxysilane-organic block copolymer compns. for manufacture of
   elec. insulating porous silica film in semiconductor device)
7631-86-9 HCAPLUS
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o = si = o

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L39 ANSWER 17 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
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AN 2003:239915 HCAPLUS

DN 138:264107

ED Entered STN: 28 Mar 2003

TI Coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device

IN Hanahata, Hiroyuki; Ioka, Takaaki

PA Asahi Kasei Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 18 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09D183-04

ICS C09D171-00; C09D183-02; C09D183-14; H01L021-316

KIND

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 37

FAN. CNT 1

FAIENT NO.		KIND	DAIL	APPLICATION NO.	DAIE	
	PI JP 200308976	59	A2	20030328	JP 2001-283539	20010918
	PRAI JP 2001-283539			20010918		
	CLASS					
	PATENT NO.	CLASS	PATENT	FAMILY CLASS	SIFICATION CODES	
		<del>-</del>				
	JP 2003089769	TCM	C09D183	8 - 04		

ICS C09D171-00; C09D183-02; C09D183-14; H01L021-316

שתעת

AB The composition comprises (A) silica precursors containing R1nSi(0R2)4-n (R1, R2 = monovalent organic residue; n = 0-3) and/or R3m(R40)3-mSiR7pSi(OR5)3-qR6q (R3-R6 = monovalent organic residue; m, q = 0-2; R7 = 0, (CH2)r; r = 1-6; p = 0, 1), their hydrolyzates, and/or their polycondensation products to satisfy Si molar fraction of mono-, di- and trifunctional alkoxysilanes to total Si in mono-, di-, tri-, tetra-, penta- and hexa-functional alkoxysilanes 1-50 mol\*, (B) linear or branched polyether block copolymer-containing organic polymers, and (C) organic solvents of alcs., ketones, amides, and/or esters. The porous insulating film claimed is obtained by coating a substrate with the composition, forming a silica/organic polymer composite film by gelation of the silica precursors, and removing the organic polymers from the composite film. The porous film has low specific dielec. constant and high durability in chemical mech. polishing in Cu wiring process for semiconductor device fabrication and scarcely generates pollutant gases in via-hole formation.

ADDITCATION NO

DAME

ST semiconductor device wiring porous silica insulator film; silica precursor org polymer coating insulator manuf; alkoxysilane polyether block copolymer coating silica film manuf

IT Silanes

RL: RCT (Reactant); RACT (Reactant or reagent)
(alkoxy; alkoxysilane- and organic polymer-containing coating composition
for manufacture of porous insulating silica film in wiring
structure in semiconductor device)

IT Dielectric films
Semiconductor devices
Sol-gel processing

(alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device) Polyoxyalkylenes, processes

RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
(block; alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Porous materials

(films; alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Films

IT

(porous; alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Polysiloxanes, preparation

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(silicate-, silica precursor; alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 24991-55-7, Polyethylene glycol dimethyl ether 106392-12-5,
Polyethylene glycol-polypropylene glycol block copolymer
RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
(alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 7631-86-9P, Silica, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 502612-69-3P, Bis(triethoxysilyl)ethane-dimethyldiethoxysilane-tetraethoxysilane copolymer

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(silica precursor; alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 106392-12-5, Polyethylene glycol-polypropylene glycol block copolymer

RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

RN 106392-12-5 HCAPLUS

CN Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME)

CM 1

CRN 75-56-9 CMF C3 H6 O

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CH<sub>3</sub>
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CM 2

CRN 75-21-8 CMF C2 H4 O

IT **7631-86-9P**, **Silica**, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (alkoxysilane- and organic polymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

RN 7631-86-9 HCAPLUS

CN Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

o = si = o

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L39 ANSWER 18 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
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AN 2003:239914 HCAPLUS

DN 138:264106

ED Entered STN: 28 Mar 2003

TI Coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device

IN Hanahata, Hiroyuki; Ioka, Takaaki

PA Asahi Kasei Corporation, Japan

SO Jpn. Kokai Tokkyo Koho, 17 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09D183-02

ICS C09D001-00; C09D005-25; C09D183-04; C09D201-00; H01L021-312

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 37

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	<b>-</b>			
PI JP 2003089768	A2	20030328	JP 2001-283571	20010918
PRAI JP 2001-283571		20010918		

CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

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JP 2003089768 ICM C09D183-02

AB The composition comprises (A) silica precursors containing (1) Si(OR1)4, their hydrolyzates, and/or their polycondensation products and

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(2) R22Si(OR1)2 and/or R23Si(OR1), their hydrolyzates, and/or their
polycondensation products (R1, R2 = monovalent organic residue) to satisfy Si
molar fraction in (2) to total Si in (1) and (2) 1-50 mol%, (B) linear or
branched block copolymer-containing organic polymers, and (C) solvents of alcs.,
ketones, amides, and/or esters. The porous insulating film is manufactured by
coating a substrate with the composition, forming a silica
/organic polymer composite film by gelation of the silica
precursors, and removing the organic polymers from the composite film.
porous film has low specific dielec. constant and high durability in
chemical mech. polishing in Cu wiring process for
semiconductor device fabrication and scarcely generates pollutant gases in
via-hole formation.
semiconductor device wiring porous silica insulator film;
silica precursor org polymer coating insulator manuf; alkoxysilane
block copolymer coating silica film manuf
Silanes
RL: RCT (Reactant); RACT (Reactant or reagent)
   (alkoxy; alkoxysilane- and block copolymer-containing coating compn
```

structure in semiconductor device) IT Dielectric films

Semiconductor devices

Sol-gel processing

(alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Polyoxyalkylenes, processes

RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (block; alkoxysilane- and block copolymer-containing coating compn . for manufacture of porous insulating silica film in wiring

structure in semiconductor device)

. for manufacture of porous insulating silica film in wiring

IT Porous materials

(films; alkoxysilane- and block copolymer-containing coating compn . for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Films

ST

IT

(porous; alkoxysilane- and block copolymer-containing coating compn. for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT Polysiloxanes, preparation

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(silicate-, silica precursor; alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 106392-12-5, Polyethylene glycol-polypropylene glycol block
copolymer

RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 7631-86-9P, Silica, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device)

IT 88029-68-9P, Dimethyldiethoxysilane-tetraethoxysilane copolymer RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (silica precursor; alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device) 78-10-4, Tetraethoxysilane 1825-62-3, Trimethylethoxysilane IT RL: RCT (Reactant); RACT (Reactant or reagent) (silica precursor; alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device) IT 106392-12-5, Polyethylene glycol-polypropylene glycol block copolymer RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses) (alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device) RN 106392-12-5 HCAPLUS CN Oxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME) CM 1 CRN 75-56-9 CMF C3 H6 O CH3 2 CM CRN 75-21-8 CMF C2 H4 O IT 7631-86-9P, Silica, uses RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (alkoxysilane- and block copolymer-containing coating composition for manufacture of porous insulating silica film in wiring structure in semiconductor device) 7631-86-9 HCAPLUS RN CN Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME) o = si = oL39 ANSWER 19 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN AN 2002:570697 HCAPLUS DN 137:133148

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ED
     Entered STN: 01 Aug 2002
     Reduction of surface roughness during chemical mechanical planarization (
ΤI
     CMP) in electronic device fabrication
IN
     Kramer, Stephen J.; Meikle, Scott G.
PA
     Micron Technology, Inc., USA
SO
     U.S., 14 pp., Cont.-in-part of U.S. Ser. No. 252,022.
     CODEN: USXXAM
DT
     Patent
LΑ
     English
IC
     ICM H01L021-00
INCL 438692000
     76-3 (Electric Phenomena)
     Section cross-reference(s): 66
FAN.CNT 2
     PATENT NO.
                       KIND
                               DATE
                                          APPLICATION NO.
                                                                  DATE
                        - - - -
     -----
                               -----
ΡI
    US 6426295
                        B1
                               20020730
                                         US 2000-584468
                                                                  20000531
    US 6409936
                        B1
                               20020625
                                           US 1999-252022
                                                                  19990216
                                           US 2002-209035
    US 2002182868
                        A1
                               20021205
                                                                  20020730
     US 6630403
                         B2
                               20031007
PRAI US 1999-252022
                        A2
                               19990216
    US 2000-584468
                        A1
                               20000531
CLASS
 PATENT NO.
                CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
                ----
 US 6426295
                ICM
                       H01L021-00
                INCL
US 6426295
                NCL
                       438/692.000; 216/038.000; 216/088.000; 216/089.000;
                       252/079.100; 438/693.000; 438/745.000
                ECLA
                       B24B037/04I; B24B057/02; C09G001/02
 US 6409936
                NCL
                       252/079.100; 216/088.000; 216/089.000; 216/090.000;
                       216/091.000; 216/092.000; 252/079.200; 252/079.300;
                       252/079.400; 252/079.500; 252/363.500; 438/690.000;
                       438/691.000; 438/692.000; 438/693.000; 451/285.000;
                       451/286.000; 451/287.000; 451/288.000
                ECLA
                       B24B037/04I; B24B057/02; C09G001/02
US 2002182868
                NCL
                       438/692.000
                ECLA
                       B24B037/04I; B24B057/02; C09G001/02
    Improved methods, compns. and structures formed therefrom are
AB
    provided that allow for reduction of roughness in layers (e.g., oxide layers)
    of a planarized wafer. In one such embodiment, improved methods,
    compns. and structures formed therefrom for reduction of roughness in
    layers (e.g., oxide layers) of a planarized wafer were used in conjunction
    with high modulus polyurethane pads. In one embodiment, improved methods,
    compns. and structures formed therefrom are provided that reduce
    rough interlayer dielec. (ILD) conditions for a wafer during CMP
    processing of such a wafer. Accordingly, this reduction of rough ILD reduces
    chatter scratches which are scratches that emanate from regions of a wafer
    that has undergone CMP processing. Advantageously, reduction in
    chatter scratching reduces cracking (i.e., wormholing) in layers of the
    wafer that were planarized. Therefore, reduction in cracking decreases access
    of cleaning chemistries to underlying structures of the wafer during
    subsequent chemical cleaning of the planarized wafer, thereby reducing damage
    to such underlying structures from these cleaning chemistries (e.g., reduction
    of metal voids in underlying metal structures). Embodiments of a method
    for forming a microelectronic substrate include mixing a surfactant at
    least 100 ppm to slurries to form a polishing solution
    The method also includes chemical-mech. planarizing of the semiconductor
    wafer using the polishing solution Addnl., embodiments of a
    polishing solution for chemical-mech. planarizing a microelectronic
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substrate includes slurries and a surfactant at least 100 ppm to
     the slurries.
ST
     chem mech polishing slurry
     surfactant electronic device fabrication
IT
        (chemical-mech.; reduction of surface roughness during
        chemical mech. planarization (CMP) in electronic device
        fabrication)
IT
     Surfactants
        (nonionic; reduction of surface roughness during chemical mech.
        planarization (CMP) in electronic device fabrication)
IT
     Polymers, uses
     Polyurethanes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing pads; reduction of surface roughness during chemical
        mech. planarization (CMP) in electronic device fabrication)
IT
     Crack (fracture)
     Dielectric films
       Slurries
     Surface roughness
     Surfactants
        (reduction of surface roughness during chemical mech. planarization (
        CMP) in electronic device fabrication)
     7631-86-9, Silica, uses 9002-92-0, Polyoxyethylene
IT
     lauryl ether
                    9004-95-9, Polyoxyethylene cetyl ether
                                  9005-00-9, Polyoxyethylene stearyl ether
     Polyoxyethylene oleyl ether
     RL: NUU (Other use, unclassified); USES (Uses)
        (reduction of surface roughness during chemical mech. planarization (
        CMP) in electronic device fabrication)
RE.CNT 16
              THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD .
RE
(1) Anon; McGraw-Hill Concise Encyclopedia of Science& Technology, Fourth
    Edition 1998, P1931
(2) Blackwell; US 5320706 A 1994
(3) Cook; US 5489233 A 1996
(4) Griesshammer; US 4070797 A 1978
(5) Griesshammer; US 4156619 A 1979
(6) Grover; US 5759917 A 1998 HCAPLUS
(7) Hiemenz, P; Principles of Colloid and Surface Chemistry, 2nd ed 1986, P428
(8) Hosali; US 5738800 A 1998
(9) Huynh; US 5704987 A 1998
(10) Isobe; US 5616212 A 1997 HCAPLUS
(11) Lyons; US 5930645 A 1999 HCAPLUS
(12) Olmstead; US 5193316 A 1993
(13) Shamouillan; US 5584146 A 1996
(14) Sherman; US 4724042 A 1988 HCAPLUS
(15) Sova; US 4563257 A 1986 HCAPLUS
(16) Wang; US 6046112 A 2000 HCAPLUS
IT
     7631-86-9, Silica, uses
     RL: NUU (Other use, unclassified); USES (Uses)
        (reduction of surface roughness during chemical mech. planarization (
        CMP) in electronic device fabrication)
RN
     7631-86-9 HCAPLUS
```

CN

Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

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L39
    ANSWER 20 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
     2002:397875 HCAPLUS
AN
DN
     136:394390
ED
    Entered STN: 28 May 2002
TT
     Chemical mechanical polishing slurries with balanced
    high polishing speed and low erosion, and manufacture of
     semiconductor devices using them in damascecne process
IN
    Nanpuku, Manabu; Yano, Hiroyuki
PA
     Toshiba Corp., Japan
SO
     Jpn. Kokai Tokkyo Koho, 7 pp.
     CODEN: JKXXAF
DT
    Patent
LΑ
    Japanese
IC
     ICM C09K003-14
     ICS C09K003-14; B24B037-00; H01L021-304; H01L021-306
CC
     76-3 (Electric Phenomena)
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                         APPLICATION NO.
                                                               DATE
     -----
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                               -----
                                          -----
    JP 2002155268
                        A2
                               20020528
                                          JP 2000-352451
                                                               20001120
PRAI JP 2000-352451
                              20001120
CLASS
 PATENT NO.
               CLASS PATENT FAMILY CLASSIFICATION CODES
                ----
 JP 2002155268 ICM
                       C09K003-14
                ICS
                       C09K003-14; B24B037-00; H01L021-304; H01L021-306
AB
    The slurry comprises a solvent, abrasives, a 1st
    surfactant, and a 2nd surfactant. The
    surfactants are preferably nonionic or those with the
    same polarity as each other, and the abrasives are preferably
    nonionic or those with the same polarity as the surfactants.
ST
    chem mech polishing slurry semiconductor damascecne;
    abrasive slurry surfactant CMP reduced erosion
IT
    Complexing agents
    Oxidizing agents
    Semiconductor device fabrication
    Surfactants
        (CMP slurry with balanced high polishing
       speed and low erosion for damascecne processes in semiconductor device
       fabrication)
IT
    Polishing materials
        (abrasive pastes; CMP slurry with
       balanced high polishing speed and low erosion for damascecne
       processes in semiconductor device fabrication)
IT
    Polyoxyalkylenes, uses
    RL: NUU (Other use, unclassified); USES (Uses)
        (alkyl allyl ethers, alkyl amines, surfactant; CMP
       slurry with balanced high polishing speed and low
       erosion for damascecne processes in semiconductor device fabrication)
IT
    Sulfates, uses
    RL: NUU (Other use, unclassified); USES (Uses)
        (alkyl derivs., surfactant; CMP slurry with
       balanced high polishing speed and low erosion for damascecne
       processes in semiconductor device fabrication)
    Quaternary ammonium compounds, uses
IT
    RL: NUU (Other use, unclassified); USES (Uses)
        (alkyltrimethyl, surfactant; CMP slurry with
       balanced high polishing speed and low erosion for damascecne
       processes in semiconductor device fabrication)
IT
    Polishing
```

(chemical-mech.; CMP slurry with balanced high polishing speed and low erosion for damascecne processes in semiconductor device fabrication)

IT Abrasives

(polishing pastes; CMP slurry with balanced high polishing speed and low erosion for damascecne processes in semiconductor device fabrication)

IT Amines, uses

RL: NUU (Other use, unclassified); USES (Uses)
(salts, alkyl derivs., surfactant; CMP slurry with
balanced high polishing speed and low erosion for damascecne
processes in semiconductor device fabrication)

IT Fatty acids, uses

RL: NUU (Other use, unclassified); USES (Uses)
(salts, surfactant; CMP slurry with balanced high
polishing speed and low erosion for damascecne processes in
semiconductor device fabrication)

IT Carbohydrates, uses

RL: NUU (Other use, unclassified); USES (Uses)
(sugar esters, surfactant; CMP slurry with balanced
high polishing speed and low erosion for damascecne processes
in semiconductor device fabrication)

IT 56-40-6, Glycine, uses 56-41-7, Alanine, uses 59-67-6, Nicotinic acid, uses 73-22-3, Tryptophane, uses 98-98-6, Picolinic acid 110-15-6, Succinic acid, uses 141-82-2, Malonic acid, uses 144-62-7, Oxalic acid, uses

RL: NUU (Other use, unclassified); USES (Uses)
(additive; CMP slurry with balanced high
polishing speed and low erosion for damascecne processes in
semiconductor device fabrication)

IT 14265-44-2, Phosphate, uses

IT

RL: NUU (Other use, unclassified); USES (Uses)
(alkyl derivs., surfactant; CMP slurry with
balanced high polishing speed and low erosion for damascecne
processes in semiconductor device fabrication)

IT 89-00-9, Quinolinic acid 93-10-7, Quinaldic acid RL: NUU (Other use, unclassified); USES (Uses)

(complexing agent; CMP slurry with balanced high polishing speed and low erosion for damascecne processes in semiconductor device fabrication)

TT 7722-84-1, Hydrogen peroxide, uses 7727-21-1, Potassium persulfate
7727-54-0, Ammonium persulfate 10421-48-4, Ferric nitrate 15078-94-1,
Ammonium cerium nitrate

RL: NUU (Other use, unclassified); USES (Uses)
 (oxidant; CMP slurry with balanced high
 polishing speed and low erosion for damascecne processes in
 semiconductor device fabrication)

56-81-5D, Glycerin, esters 1331-61-9, Dodecylbenzenesulfonic acid ammonium salt 1875-92-9D, Dimethylbenzyl ammonium chloride, alkyl derivs. 9002-92-0, Poly(oxyethylene) lauryl ether 12441-09-7D, Sorbitan, esters 25322-68-3D, alkyl allyl ethers, alkyl amines

27177-77-1, Dodecylbenzenesulfonic acid potassium salt Naphthalenesulfonic acid potassium salt, alkyl derivs.

RL: NUU (Other use, unclassified); USES (Uses) (surfactant; CMP slurry with balanced high polishing speed and low erosion for damascecne processes in

semiconductor device fabrication)

L39 ANSWER 21 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN AN 2002-148168 [19] WPIX

94653-96-0D,

```
C2002-046084
DNC
TI
     Chemical mechanical polishing slurry
     composition for polishing substrate comprises
     abrasive and oxidizing agent.
DC
     G04 P61
IN
     MAHULIKAR, D; PASQUALONI, A M
PA
     (ARCH-N) ARCH SPECIALTY CHEM INC; (PLAN-N) PLANAR SOLUTIONS LLC
CYC
     25
PΙ
     WO 2002004573
                     A2 20020117 (200219)* EN
                                                18
                                                      C09K000-00
        RW: AT BE CH CY DE DK ES FI FR GB GR IE IT LU MC NL PT SE TR
         W: JP KR SG
     US 6468913
                     B1 20021022 (200273)
                                                      H01L021-00
     EP 1354017
                     A2 20031022 (200370)
                                          EN
                                                      C09K003-00
         R: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE TR
     KR 2003059070
                   A 20030707 (200377)
                                                      C09K003-14
     JP 2004502860
                     W 20040129 (200413)
                                                32
                                                      C09K003-14
     TW 574347
                     A 20040201 (200453)
                                                      C09K013-00
ADT
    WO 2002004573 A2 WO 2001-US10491 20010402; US 6468913 B1 US 2000-611702
     20000708; EP 1354017 A2 EP 2001-922992 20010402, WO 2001-US10491 20010402;
     KR 2003059070 A KR 2003-700216 20030107; JP 2004502860 W WO 2001-US10491
     20010402, JP 2002-509430 20010402; TW 574347 A TW 2001-113232 20010531
FDT EP 1354017 A2 Based on WO 2002004573; JP 2004502860 W Based on WO
     2002004573
PRAI US 2000-611702
                          20000708
     ICM C09K000-00; C09K003-00; C09K003-14; C09K013-00; H01L021-00
IC
     ICS B24B037-00; C09G001-00; H01L021-304
AB
     WO 200204573 A UPAB: 20040920
     NOVELTY - Chemical mechanical polishing
     slurry composition comprises an abrasive
     including silica, alumina, ceria, or their mixtures; and an
     oxidizing agent. The slurry has an effective shelf life of at
     least 30 days.
          DETAILED DESCRIPTION - Chemical mechanical
     polishing slurry composition comprises an
     abrasive including silica, alumina, ceria, or their
     mixtures; and an oxidizing agent including hydrogen peroxide, potassium
     ferricyanide, potassium dichromate, potassium iodate, potassium bromate,
     vanadium trioxide, hypochlorous acid, sodium hypochlorite, potassium
     hypochlorite, calcium hypochlorite, magnesium hypochlorite, ferric
     nitrate, ammonium persulfate, potassium permanganate, or their mixtures.
     The slurry has an effective shelf life of at least 30 days.
     INDEPENDENT CLAIMS are also included for (A) a method of polishing
     a substrate, which comprises providing a substrate with at least one metal.
     layer, applying the inventive slurry composition, and
     chemically mechanically polishing the
     substrate with the slurry; and (B) a method of preparing a
     chemical mechanical polishing slurry
     , which comprises mixing an abrasive with an oxidizer, and
     storing the slurry prior to use.
         USE - The slurry composition is used for
     chemical mechanical polishing of metal
     substrates on semiconductor wafers.
         ADVANTAGE - The inventive slurry composition is
     ready-to-use and stable over long term storage at ambient conditions. Its
    use in a chemical mechanical polishing
    process promotes high removal rates, low defect densities, and reduced
     amounts of dishing and erosion.
    Dwg.0/0
FS
    CPI GMPI
FA
     AB
```

```
MC
     CPI: G04-B04
    ANSWER 22 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
L39
AN
     2003-185553 [19]
                       WPIX
DNN
    N2003-146128
                       DNC C2003-049112
ΤI
     Chemical/mechanical polishing slurry
     , used in producing shallow trench insulation in silicon wafer with oxide
     and nitride coatings, comprises abrasive particles in aqueous
     solution containing two different passivating agents.
DC
     A85 E19 G02 L02 L03 P61 U11
    HAH, S; LEE, J; YOON, B; HA, S R; LEE, J D; LEE, J W; YOON, B E
IN
PA
     (SMSU) SAMSUNG ELECTRONICS CO LTD; (HAHS-I) HAH S; (LEEJ-I) LEE J;
     (YOON-I) YOON B
CYC
    6
    DE 10160174
                    A1 20021017 (200319) *
                                                      C09G001-02
PΙ
                                                16
     JP 2002313760
                    A 20021025 (200319)
                                                11
                                                      H01L021-304
    KR 2002077636 A 20021012 (200319)
                                                     H01L021-304
    US 2003022499 A1 20030130 (200319)
                                                     H01L021-302
                    B2 20030401 (200324)
    US 6540935
                                                     C09K013-00
    US 2003148616 A1 20030807 (200358)
                                                     C03C025-68
                    A1 20040326 (200427)
    SG 102648
                                                      C09K013-00
    TW 573001
                    A 20040121 (200453)
                                                      C09K003-14
    KR 459696
                    B 20041203 (200525)
                                                     H01L021-304
    US 6914001
                    B2 20050705 (200544)
                                                     H01L021-302
ADT DE 10160174 A1 DE 2001-10160174 20011207; JP 2002313760 A JP 2001-392645
     20011225; KR 2002077636 A KR 2001-25873 20010511; US 2003022499 A1 US
     2001-826169 20010405; US 6540935 B2 US 2001-826169 20010405; US 2003148616
    Al Div ex US 2001-826169 20010405, US 2003-351539 20030127; SG 102648 A1
     SG 2001-7788 20011212; TW 573001 A TW 2001-129490 20011129; KR 459696 B KR
     2001-25873 20010511; US 6914001 B2 Div ex US 2001-826169 20010405, US
     2003-351539 20030127
FDT
    US 2003148616 Al Div ex US 6540935; KR 459696 B Previous Publ. KR
     2002077636; US 6914001 B2 Div ex US 6540935
PRAI US 2001-826169
                         20010405; US 2003-351539
    ICM C03C025-68; C09G001-02; C09K003-14; C09K013-00; H01L021-302;
         H01L021-304
     ICS B24B037-00; B24B057-02; H01L021-461
AΒ
    DE 10160174 A UPAB: 20030320
    NOVELTY - Chemical/mechanical polishing
     slurry comprises an aqueous solution containing abrasive
    particles and 2 different passivating agents.
         USE - The slurry is used in rotary chemical/
    mechanical polishing (CMP) with a
    polishing pad; and in shallow trench insulation, comprising
    coating a semiconductor substrate with cushion oxide and silicon nitride
     (SiN), making trenches through these layers into the substrate, forming an
    insulating oxide film over the trenches and removing the SiN layer and
    this film down to the plane of the SiN layer by CMP (all
    claimed). The CMP is used in making microelectronic devices.
         ADVANTAGE - The oxide/silicon nitride selectivity of the aqueous
    solution is not less than 50:1 (claimed), which is better than usual.
         DESCRIPTION OF DRAWING(S) - The drawing shows the chemical/
    mechanical polishing (CMP) stage of the
    process flow diagram. (Drawing includes non-English language text).
         Contact surface of wafer with polishing pad 802
         Supply CMP slurry containing abrasive
    and the first and second passivating agents 803
         Rotate surface of wafer relative to surface of polishing
    pad to remove oxide coating 804
    Dwg.8/9
```

Polishing materials

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FS
    CPI EPI GMPI
FΑ
    AB; GI; DCN
    CPI: A11-B05; A12-E04; A12-E07C; E05-A; E05-G09C; E10-A09B4; E10-A09B5;
MC
         E10-A22; E10-C02F; E10-C04; E31-K05D; E31-K05E; E31-P03; E31-P06D;
         E34-C02; E34-E; E35-G; E35-K02; E35-L; G02-A05B; L02-F; L04-C27
    EPI: U11-A10; U11-C06A1A
    ANSWER 23 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
    2001:338649 HCAPLUS
AN
DN
    134:341709
    Entered STN: 11 May 2001
ED
TI
    Organic additives for chemical-mechanical polishing
    slurries for Ta barrier layers in integrated circuits
    Sahota, Kashmir S.; Schonauer, Diana M.; Avanzino, Steven C.
TN
    Advanced Micro Devices, Inc., USA
PA
    PCT Int. Appl., 38 pp.
SO
    CODEN: PIXXD2
DT
    Patent
    English
LA
IC
    ICM C09G001-02
     42-13 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 76
FAN.CNT 3
    PATENT NO.
                                        APPLICATION NO.
                       KIND
                             DATE
                                                              DATE
     ------
                              _____
                                         -----
    WO 2001032794
                        A1
                              20010510 WO 2000-US30354
PΙ
                                                               20001103
        W: CN, JP, KR, SG
        RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
            PT, SE, TR
    US 2002005504
                        Α1
                              20020117
                                         US 1999-434146
                                                            19991104
    US 6503418
                        B2
                              20030107
                                       EP 2000-976902
    EP 1246879
                              20021009
                        A1
                                                               20001103
           AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, FI, CY, TR
    JP 2003514374
                        T2
                              20030415
                                         JP 2001-535479
                                                               20001103
PRAI US 1999-434146
                        Α
                              19991104
    WO 2000-US30354
                        W
                              20001103
CLASS
 PATENT NO.
               CLASS PATENT FAMILY CLASSIFICATION CODES
 -----
               _____
WO 2001032794
               ICM
                      C09G001-02
WO 2001032794
               ECLA
                      C09G001/02; C09K003/14D2
US 2002005504
               NCL
                      252/079.100
                ECLA C09G001/02; C09K003/14D2
AΒ
    A Ta barrier slurry for Chemical-Mech. Polishing (
    CMP) during copper metalization contains an organic additive which
    suppresses formation of ppts. and copper staining. The organic additive is
    chosen from a class of compds. which form multiple strong adsorbent bonds
    to the surface of silica or copper, which provide a high degree of surface
    coverage onto the reactive species, thereby occupying potential reaction
    sites, and which are sized to sterically hinder the collisions between two
    reactant mols. which result in new bond formation.
ST
    org additive polishing slurry integrated circuit
ΙT
    Surfactants
       (nonionic; organic additives for chemical-mech. polishing
       slurries for Ta barrier layers in integrated circuits)
IT
    Abrasives
    Corrosion inhibitors
    Integrated circuits
```

```
(organic additives for chemical-mech. polishing slurries
       for Ta barrier layers in integrated circuits)
IT
     Amines, uses
     Polyoxyalkylenes, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (organic additives for chemical-mech. polishing slurries
       for Ta barrier layers in integrated circuits)
     7631-86-9, Silica, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (abrasive; organic additives for chemical-mech. polishing
       slurries for Ta barrier layers in integrated circuits)
IT
     50-70-4, Sorbitol, uses 56-81-5, Glycerol, uses 107-21-1, Ethylene
    glycol, uses 111-46-6, Di(ethylene glycol), uses 9002-89-5, Polyvinyl
     alcohol
              9002-92-0, Polyoxyethylene lauryl ether 9003-05-8,
     Polyacrylamide 9004-95-9, Polyoxyethylene cetyl ether 9016-45-9,
    Nonylphenol polyethylene oxide 9063-89-2, Poly(ethylene oxide)
    octylphenol ether 25067-34-9, Ethylene-vinyl alcohol copolymer
     25213-24-5, Vinyl alcohol-vinyl acetate copolymer 25322-68-3,
     Poly(ethylene glycol) 25791-96-2
                                      31694-55-0, Polyethylene glycol
    glycerol ether 156048-32-7, Dimethylsilanediol-ethylene oxide co-polymer
    RL: MOA (Modifier or additive use); USES (Uses)
        (organic additives for chemical-mech. polishing slurries
       for Ta barrier layers in integrated circuits)
RE.CNT 2
             THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE:
(1) Gomez, J; US 5897375 A 1999
(2) Higuchi, M; US 5770095 A 1998 HCAPLUS
    ANSWER 24 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
L39
AN
    2001:469560 HCAPLUS
DN
    135:63058
ED
    Entered STN: 29 Jun 2001
ΤI
    Cleaning compositions for chemical mechanical
    polishing slurry of semiconductors
IN
    Yamana, Masahide; Shudo, Shinsei
PΑ
    Advantech K. K., Japan
so
    Jpn. Kokai Tokkyo Koho, 5 pp.
    CODEN: JKXXAF
DΤ
    Patent
LA
    Japanese
IC
    ICM H01L021-304
    ICS B24B037-00; C11D001-00; C11D003-14
CC
    46-6 (Surface Active Agents and Detergents)
    Section cross-reference(s): 76
FAN.CNT 1
    PATENT NO.
                                    APPLICATION NO.
                       KIND DATE
                                                             DATE
    ------
                                         -----
                       ----
                                                               -----
    JP 2001176825
                      A2
PΙ
                              20010629
                                         JP 1999-356160
                                                               19991215
PRAI JP 1999-356160
                             19991215
CLASS
PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
 -----
               JP 2001176825 ICM H01L021-304
               ICS B24B037-00; C11D001-00; C11D003-14
AB
    The title compns. contain 1-70% abrasive particles
    free from alkali metal ions (e.g., silica with average diameter 25
    μm) and 1-50% nonionic surfactants (e.g.,
    polyoxyethylene monooleate, polyoxyethylene propylene glycol,
    polyoxyethylene monostearate) as aqueous emulsions.
ST
    semiconductor chem mech polishing
```

GB 2354525

B2

20031022

```
slurry cleaning compn; silica abrasive
      particle semiconductor polishing slurry cleaning;
      nonionic surfactant semiconductor polishing
      slurry cleaning
      Emulsions
 .IT
      Scouring agents
      Semiconductor materials
         (cleaning compns. for chemical mech.
         polishing slurry of semiconductors)
      Surfactants
 IT
         (nonionic; cleaning compns. for chemical
         mech. polishing slurry of semiconductors)
 IT
         (of semiconductors; cleaning compns. for chemical
         mech. polishing slurry of semiconductors)
 TT
      Abrasives
         (particles; cleaning compns. for chemical mech
         . polishing slurry of semiconductors)
 IT
      7631-86-9, Silica, uses
      RL: TEM (Technical or engineered material use); USES (Uses)
         (abrasive particles; cleaning compns. for
         chemical mech. polishing slurry of
         semiconductors)
      9003-11-6, Polyoxyethylene propylene glycol ether 9004-96-0,
· IT
      Polyoxyethylene monooleate 9004-99-3, Polyoxyethylene monostearate
      RL: TEM (Technical or engineered material use); USES (Uses)
         (nonionic surfactants; cleaning compns.
         for chemical mech. polishing slurry
         of semiconductors)
 IT
      7631-86-9, Silica, uses
      RL: TEM (Technical or engineered material use); USES (Uses)
         (abrasive particles; cleaning compns. for
         chemical mech. polishing slurry of
         semiconductors)
 RN
      7631-86-9 HCAPLUS
 CN
      Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)
 o = si = o
 L39
      ANSWER 25 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
 AN
      2001:557817 HCAPLUS
 DN
      135:115843
 ED
      Entered STN: 03 Aug 2001
 ΤI
      Polishing compositions for magnetic disks
 IN
      Shemo, David M.; Rader, W. Scott; Owaki, Toshiki
      Fujimi America Inc., USA
 PA
 SO
      Brit. UK Pat. Appl., 36 pp.
      CODEN: BAXXDU
 DT
      Patent
 LΑ
      English
 IC
      ICM C09G001-02
 CC
      77-8 (Magnetic Phenomena)
 FAN.CNT 1
      PATENT NO.
                          KIND
                                 DATE
                                             APPLICATION NO.
                                                                    DATE
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 ΡI
      GB 2354525
                           A1
                                 20010328
                                             GB 2000-23101
                                                                    20000920
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US 1999-404993
     US 6258140
                         B1
                               20010710
                                                                   19990927
     CN 1289811
                         Α
                               20010404
                                           CN 2000-131713
                                                                  20000927
                               20040114
     CN 1134521
                         В
                         A2
     JP 2001155332
                               20010608
                                           JP 2000-294874
                                                                  20000927
PRAI US 1999-404993
                         Α
                               19990927
CLASS
 PATENT NO.
                CLASS PATENT FAMILY CLASSIFICATION CODES
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                       __________
              ICM
 GB 2354525
                       C09G001-02
 GB 2354525
                ECLA
                       C09G001/02; C09K003/14D2; G11B005/84B
 US 6258140
                NCL
                        051/308.000; 106/003.000; 510/165.000; 510/167.000
                 ECLA
                        C09G001/02; C09K003/14D2; G11B005/84B
AB
     A polishing composition for polishing a memory
    hard disk having a high stock removal rate and capable of providing a
    polished surface having a small surface roughness and preventing
     formation of microprotrusions, micropits or other surface defects,
     comprises the following components: (a) 0.1-50 wt% in the total amount of
     the polishing composition, of \geq 1 abrasive
     particles selected from silicon dioxide, aluminum oxide, cerium oxide,
     zirconium oxide, titanium oxide, silicon nitride and manganese dioxide;
     (b) 0.0001-3.0 wt% of ≥1 polishing resistance-reducing
     agent selected from a surfactant, a water-soluble polymer and a
     polyelectrolyte, (c) 0.001-40 wt% of \geq 1 polishing
     accelerator selected from an inorg. acid, an organic acid and their aluminum,
     iron, nickel and cobalt salts (such as aluminum nitrate, ascorbic acid,
     sulfuric acid), and (d) water.
ST
     magnetic disk polishing compn; silicon dioxide
     abrasive polishing; polishing resistance
     reducing agent; accelerator polishing
IT
     Salts, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (Al, Fe, Ni, Co; polishing compns. for magnetic
        disks)
IT
     Acrylic polymers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (ammonium salt; polishing compns. for magnetic
        disks)
IT
     Surfactants
        (anionic; polishing compns. for magnetic disks)
IT
    Surfactants
        (cationic; polishing compns. for magnetic disks)
IT
    Amines, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
        (coco alkyl, ethoxylated, quaternary ammonium chloride;
       polishing compns. for magnetic disks)
ΙT
    Amines, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
        (ethoxylated, quaternary ammonium salt; polishing
        compns. for magnetic disks)
IT
     Surfactants
        (nonionic; polishing compns. for magnetic
        disks)
IT
    Sulfonic acids, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing accelerator; polishing compns.
        for magnetic disks)
IT
    Abrasives
    Magnetic disks
      Polishing
      Polishing materials
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GOUDREAU 10/807139 09/16/2005

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Polyelectrolytes
       Slurries
        (polishing compns. for magnetic disks)
IT
     Polyoxyalkylenes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing compns. for magnetic disks)
     Amines, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (tallow alkyl, ethoxylated; polishing compns. for
        magnetic disks)
     7631-86-9, Colloidal silica, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (colloidal; polishing compns. for magnetic disks)
IT
     50-21-5, Lactic acid, uses 50-81-7, Ascorbic acid, uses
     Glycine, uses
                   56-86-0, Glutamic acid, uses
                                                     77-92-9, Citric acid, uses
     79-14-1, Glycolic acid, uses
                                   87-69-4, Tartaric acid, uses
     Mandelic acid 110-15-6, Succinic acid, uses
                                                    141-82-2, Malonic acid,
     uses
            473-81-4, Glyceric acid 526-95-4, Gluconic acid
     Nickel oxalate
                     814-87-9, Aluminum oxalate
                                                   2338-05-8, Iron citrate
     6915-15-7, Malic acid
                            7050-19-3, Ammonium iron citrate
                                                                 7446-70-0.
                              7601-90-3, Perchloric acid, uses
     Aluminum chloride, uses
                                                                   7646-79-9,
     Cobalt chloride, uses
                           7647-01-0, Hydrochloric acid, uses
                                                                   7664-38-2,
     Phosphoric acid, uses
                             7664-93-9, Sulfuric acid, uses
                                                              7697-37-2, Nitric
                  7705-08-0, Iron chloride, uses
     acid, uses
                                                   7718-54-9, Nickel chloride,
            7784-25-0, Ammonium aluminum sulfate
                                                    7786-81-4, Nickel sulfate
     10043-01-3, Aluminum sulfate
                                   10043-35-3, Boric acid, uses
                                                                   10045-89-3,
     Ammonium iron sulfate
                             10124-43-3, Cobalt sulfate
                                                           10124-49-9, Iron
     sulfate
             10141-05-6, Cobalt nitrate
                                           13138-45-9, Nickel nitrate
                                    13637-71-3, Nickel perchlorate
     13473-90-0, Aluminum nitrate
     14013-86-6, Iron nitrate
                                14452-39-2, Aluminumperchlorate
                                                                   19297-92-8,
     Ammonium iron oxalate
                             22605-92-1, Nickel citrate
                                                           31142-56-0, Aluminum
     citrate
               38023-20-0, Iron hydroxide perchlorate (Fe(OH)(ClO4)2)
     49599-05-5, Ammonium aluminum citrate
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing accelerator; polishing compns.
        for magnetic disks)
     60-00-4D, Ethylenediaminetetraacetic acid, Fe, Ni, Co complex, uses
IT
     67-42-5D, Fe, Ni, Co complex 67-43-6D, Diethylenetriaminepentaacetic
     acid, Fe, Ni, Co complex 93-62-9D, Fe, Ni, Co complex Nitrilotriacetic acid, Fe, Ni, Co complex 150-25-4D, D
                                                                139-13-9D,
                                                150-25-4D, Dihydroxyethyl
     glycine, Fe, Ni, Co complex 150-39-0D, Hydroxyethylethylenediaminetriace
     tic acid, Fe, Ni, Co complex 869-52-3D, Fe, Ni, Co complex 1306-38-3,
     Cerium oxide, uses
                         1313-13-9, Manganese dioxide, uses
                                                               1314-23-4,
     Zirconium oxide, uses
                             1344-28-1, Aluminum oxide, uses
                                                                4408-81-5D.
     Propylenediaminetetraacetic acid, Fe, Ni, Co complex 9002-89-5,
     Polyvinyl alcohol
                         9003-01-4, Polyacrylic acid 12033-89-5, Silicon
     nitride, uses
                     13463-67-7, Titanium oxide, uses
                                                       25322-68-3,
     Polyethylene oxide
                          162362-34-7
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing compns. for magnetic disks)
L39
     ANSWER 26 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     2001-168335 [17]
                        WPTX
DNN
     N2001-121419
                        DNC C2001-050197
     Chemical-mechanical polishing slurry
TI
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- AN
- used in semiconductor device planarization has particles uniformly dispersed in an aqueous medium with specified surface area, an aggregate size distribution and an aggregate diameter.
- DC A25 A26 A85 L03 M14 P61 U11
- IN BURKE, PA; LACK, CD; LUO, Q; SACHAN, V; THOMAS, TM; YE, QC; YE, Q

MC

L39

AN

CPI: A12-W12C; L04-C26; M14-A

WPIX

EPI: U11-C06A1A

2001-158580 [16]

DNN N2001-115522

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PA
     (RODE-N) RODEL HOLDINGS INC
CYC
    31
PΙ
    WO 2001002134
                    A1 20010111 (200117)* EN
                                                33
                                                      B24B001-00
        RW: AT BE CH CY DE DK ES FI FR GB GR IE IT LU MC NL PT SE
        W: CN JP KR SG
     EP 1177068
                    A1 20020206 (200218) EN
                                                      B24B001-00
        R: AL AT BE CH CY DE DK ES FI FR GB GR IE IT LI LT LU LV MC MK NL PT
           RO SE SI
     TW 452523
                    A 20010901 (200240)
                                                      B24B001-00
    US 6447373
                    B1 20020910 (200263)
                                                      B24B001-00
     KR 2002035826
                    A 20020515 (200273)
                                                      H01L021-304
    JP 2003503862
                    W 20030128 (200309)
                                                34
                                                      H01L021-304
    WO 2001002134 A1 WO 2000-US17046 20000621; EP 1177068 A1 EP 2000-943003
     20000621, WO 2000-US17046 20000621; TW 452523 A TW 2000-112525 20000626;
     US 6447373 B1 Provisional US 1999-142326P 19990703, US 2000-598377
     20000621; KR 2002035826 A KR 2002-700019 20020102; JP 2003503862 W WO
     2000-US17046 20000621, JP 2001-507608 20000621
    EP 1177068 A1 Based on WO 2001002134; JP 2003503862 W Based on WO
     2001002134
PRAI US 1999-142326P
                          19990703; US 2000-598377
                                                         20000621
     ICM B24B001-00; H01L021-304
        B24B037-00; B24D003-34; C09K003-14; C09K013-00; H01L021-00;
         H01L021-44
AΒ
    WO 200102134 A UPAB: 20010328
    NOVELTY - Chemical-mechanical polishing
     slurry has particles uniformly dispersed in an aqueous medium with
     a surface area of 40-430 m2/g, aggregate size distribution less than 1 mu
    m, mean aggregate diameter less than 0.4 mu m, and contains at least one
     ionic species which prevents a force sufficient to repel and overcome the
     van der Waals forces between particles. The particles form stage 1
     agglomerates greater than 1 mu m.
         DETAILED DESCRIPTION - An INDEPENDENT CLAIM is also included for a
    method of chemical-mechanical polishing of a
    metal layer of a substrate by chemical mechanical
    polishing a metal layer with the inventive slurry
    comprising particles dispersible in an aqueous medium. The medium has a
     static etch rate with respect to the metal layer of less than 75 Angstrom
     /min.
          USE - The slurry is used in chemical-
    mechanical polishing of a metal layer of a substrate,
    preferably in semiconductor device planarization, memory disk
    polishing, and optics polishing. It can be used in
    polishing dielectrics, including low k (dielectric constant)
    dielectrics, e.g. porous silica, or organic low k dielectrics,
    e.g. fluoro polymers or copolymers.
         ADVANTAGE - The method provides an effective polishing to
    metal layers at desired polishing rates while minimizing surface
     imperfections and defects. The slurries have a low static etch
    rate and are metastable due to reversible formation of types of
    agglomerates. The agglomerates will not cause unacceptable
    polishing defects and will de-agglomerate with simple agitation.
    Dwg.0/6
FS
    CPI EPI GMPI
FΑ
    AB
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ANSWER 27 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN

DNC C2001-046981

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GOUDREAU 10/807139 09/16/2005
                                          Page 51
TT
     Manufacture of a semiconductor device on a wafer involves chemically
     treating the wafer by immersion in a solution and/or double sided
     scrubbing of the wafer with the solution.
DC
     L03 U11
IN
     AVANZINO, S C; SCHONAUER, D M; YANG, K
     (ADMI) ADVANCED MICRO DEVICES INC
PA
CYC
PΙ
     US 6177349
                    B1 20010123 (200116)*
                                                7
                                                     H01L021-44
     US 6177349 B1 US 1998-206169 19981207
ADT
PRAI US 1998-206169
                          19981207
IC
     ICM H01L021-44
AΒ
     US
          6177349 B UPAB: 20010323
     NOVELTY - A semiconductor device is manufactured on a wafer by chemically
     treating the wafer surface with a solution containing ammonium fluoride,
     diammonium hydrogen citrate, triammonium citrate, a surfactant, and water
     by immersing the wafer in the solution and/or double sided brush scrubbing
     the wafer with the solution.
          DETAILED DESCRIPTION - Manufacture of a semiconductor device on a
     wafer involves:
          (a) forming a copper (Cu) or Cu alloy interconnection pattern
     comprising a dense array of spaced apart Cu or Cu alloy lines bordering an
     open dielectric field on a surface of the wafer; and
          (b) chemically treating the wafer surface with a solution containing
     ammonium fluoride, diammonium hydrogen citrate, triammonium citrate, a
     surfactant, and water by immersing the wafer in the solution and/or double
     sided brush scrubbing the wafer with the solution.
          USE - Manufacturing high speed integrated circuits having submicron
     features and high aspect ratio openings such as semiconductor devices with
     a design rule of 0.18 microns and under.
          ADVANTAGE - Enables the formation of reliable Cu and/or Cu alloy
     interconnection while eliminating or substantially reducing the formation
     and/or growth of dendritics emanating from Cu or Cu alloy lines and
     substantially removing residual slurry particles in an
     efficient, cost effective manner.
     Dwg.0/1
     CPI EPI
FS
FA
     AB
MC
     CPI: L04-C11C; L04-C26
     EPI: U11-C05E1; U11-C06A1A
L39
    ANSWER 28 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
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2000:291176 HCAPLUS AN

DN 132:302004

ED Entered STN: 05 May 2000

Chemical mechanical polishing slurry system having an TIactivator solution

IN Mahulikar, Deepak

PA Arch Specialty Chemicals, Inc., USA

SO PCT Int. Appl., 21 pp.

CODEN: PIXXD2

DTPatent

LΑ English

IC ICM C09K003-14

ICS C09G001-02; B24B001-00

CC 76-3 (Electric Phenomena)

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE ---------ΡI WO 2000024842 A1 20000504 WO 1999-US24864 19991022

W: JP, KR, SG

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RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE
                               20010822
                                           EP 1999-955147
    EP 1124912
                         Α1
        R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, FI
    JP 2002528903
                         T2
                               20020903
                                           JP 2000-578398
                                                                  19991022
    US 6447563
                         B1
                               20020910
                                           US 1999-425358
PRAI US 1998-105366P ·
                         P
                               19981023
    WO 1999-US24864
                         W
                               19991022
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                 _ _ _ _ _
                        ICM
WO 2000024842
                       C09K003-14
                ICS
                       C09G001-02; B24B001-00
WO 2000024842
                ECLA
                       C09G001/02; C09K003/14D2
US 6447563
                NCL
                        051/309.000; 051/307.000; 051/308.000; 106/003.000;
                        252/079.200; 252/079.300; 252/079.400; 438/692.000;
                        438/693.000
                        C09G001/02; C09K003/14D2
    This invention relates to a CMP slurry system for use
AΒ
     in semiconductor device fabrication. The slurry system
     comprises 2 parts. The 1st part is a generic dispersion that contains
    only an abrasive and, optionally, a surfactant and a stabilizing
     agent. The generic dispersion can be used for polishing metals
    as well as interlayer dielecs. The 2nd part is a novel activator solution
     comprising ≥2 components selected from: an oxidizer, acids, amines,
     chelating agents, F-containing compds., corrosion inhibitors, buffering
    agents, surfactants, biol. agents, and their mixts.
ST
    chem mech polishing slurry activator soln;
    semiconductor device fabrication CMP slurry
IT
    Quaternary ammonium compounds, processes
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
    engineered material use); PROC (Process); USES (Uses)
        (alkylbenzyldimethyl, chlorides; chemical mech. polishing
        slurry system having activator solution containing)
IT
    Surfactants
        (amphoteric; chemical mech. polishing slurry system
       having activator solution containing)
IT
    Surfactants
        (anionic; chemical mech. polishing slurry system
       having activator solution containing)
IT
    Surfactants
        (cationic; chemical mech. polishing slurry system
       having activator solution containing)
IT
    Abrasives
    Buffers
    Chelating agents
    Corrosion inhibitors
    Oxidizing agents
    Stabilizing agents
    Surfactants
        (chemical mech. polishing slurry system having
       activator solution containing)
IT
    Acids, processes
    Alkali metal fluorides
    Alkaline earth fluorides
      Amines, processes
    Carboxylic acids, processes
    RL: PEP (Physical, engineering or chemical process); TEM (Technical or
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IT

IT

IT

IT

IT

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IT

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engineered material use); PROC (Process); USES (Uses)
   (chemical mech. polishing slurry system having
   activator solution containing)
Semiconductor device fabrication
  Slurries
   (chemical mech. polishing slurry system having
   activator solution for semiconductor device fabrication)
Polishing
   (chemical-mech.; chemical mech. polishing slurry system
   having activator solution for semiconductor device fabrication)
Electric insulators
   (interlayer; slurry system having activator solution for
   chemical-mech. polishing of)
Surfactants
   (nonionic; chemical mech. polishing slurry
   system having activator solution containing)
50-21-5, Lactic acid, processes
                                  56-34-8, Tetraethylammonium chloride
60-00-4, Ethylenediaminetetraacetic acid, processes
                                                     64-18-6, Formic
acid, processes
                 64-19-7, Acetic acid, processes
                                                    67-43-6,
Diethylenetriaminepentaacetic acid
                                    75-57-0, Tetramethylammonium chloride
75-59-2, Tetramethylammonium hydroxide 77-92-9, Citric acid,
processes
           79-09-4, Propanoic acid, processes 87-69-4, Tartaric acid,
            88-99-3, Phthalic acid, processes 95-14-7, 1H-Benzotriazole
processes
102-71-6, Triethanolamine, processes 103-76-4, 1-Piperazineethanol
103-83-3D, Benzyldimethylamine, alkyl ammonium hydroxide derivs.
107-92-6, Butanoic acid, processes 109-52-4, Pentanoic acid, processes
111-14-8, Heptanoic acid 111-42-2, Diethanolamine, processes
Nonanoic acid
               124-07-2, Octanoic acid, processes
                                                     136-85-6,
6-Tolyltriazole
                 139-13-9, Nitrilotriacetic acid
Monoethanolamine, processes
                             142-62-1, Hexanoic acid, processes
149-91-7, Gallic acid, processes
                                  150-39-0, N-
Hydroxyethylethylenediaminetriacetic acid
                                           373-68-2, Tetramethylammonium
fluoride
          409-21-2, Silicon carbide (SiC), processes
                                                        526-95-4, Gluconic
acid
       627-74-7
                 929-06-6, Diethyleneglycolamine
                                                   1306-38-3, Ceria,
processes
           1310-58-3, Potassium hydroxide, processes
                                                        1314-23-4,
Zirconium oxide, processes 1332-29-2, Tin oxide
                                                   1332-37-2, Iron oxide,
processes 1336-21-6, Ammonium hydroxide ((NH4)(OH))
                                                      1341-49-7,
Ammonium bifluoride
                      1344-28-1, Alumina, processes
                                                      3811-73-2, Sodium
pyrithione
            4499-86-9, Tetrapropylammonium hydroxide
                                                        5810-42-4,
Tetrapropylammonium chloride 6915-15-7, Malic acid
                                                       7647-01-0,
Hydrochloric acid, processes
                              7664-38-2, Phosphoric acid, processes
7664-39-3, Hydrofluoric acid, processes 7664-93-9, Sulfuric acid,
processes
           7681-52-9, Sodium hypochlorite
                                             7697-37-2, Nitric acid,
processes
           7758-19-2, Sodium chlorite 7803-49-8, Hydroxylamine,
processes
           12033-89-5, Silicon nitride, processes
                                                    12125-01-8, Ammonium
fluoride
          13463-67-7, Titanium dioxide, processes
                                                    35914-36-4,
Pyrogallol carboxylic acid
                            57178-78-6
                                          68444-11-1
                                                       123155-80-6
130397-22-7, Perfluoric acid
                              152275-68-8, 1-
(2,3,Dicarboxypropyl)benzotriazole
RL: PEP (Physical, engineering or chemical process); TEM (Technical or
engineered material use); PROC (Process); USES (Uses)
   (chemical mech. polishing slurry system having
   activator solution containing)
7631-86-9, Silica, processes
RL: PEP (Physical, engineering or chemical process); TEM (Technical or
engineered material use); PROC (Process); USES (Uses)
   (colloidal; chemical mech. polishing slurry system
  having activator solution containing)
7429-90-5, Aluminum, processes
                                7440-25-7, Tantalum, processes
7440-32-6, Titanium, processes
                                7440-33-7, Tungsten, processes
```

7440-50-8, Copper, processes

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)

(slurry system having activator solution for chemical-mech. polishing of)

RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Feller; US 5700383 A 1997 HCAPLUS
- (2) Kaufman; US 5783489 A 1998
- (3) Kido; US 5800577 A 1998 HCAPLUS
- (4) Neville; US 5527423 A 1996 HCAPLUS
- (5) Sandusky; US 5266088 A 1998 HCAPLUS
- (6) Yamada; US 5366542 A 1994 HCAPLUS
- IT 75-59-2, Tetramethylammonium hydroxide 1336-21-6,

Ammonium hydroxide ((NH4)(OH))

RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(chemical mech. polishing slurry system having

activator solution containing)

RN 75-59-2 HCAPLUS

CN Methanaminium, N,N,N-trimethyl-, hydroxide (9CI) (CA INDEX NAME)

● OH -

RN 1336-21-6 HCAPLUS

CN Ammonium hydroxide ((NH4)(OH)) (9CI) (CA INDEX NAME)

H4N-OH

L39 ANSWER 29 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN

AN 2000:803851 HCAPLUS

DN 133:358135

ED Entered STN: 15 Nov 2000

Auxiliary agents and compositions containing water-soluble polymers for chemical-mechanical polishing of semiconductor substrates or devices

IN Ishibashi, Yoichi; Sowa, Toshiki; Fukumoto, Yasuhisa

PA Kao Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08L101-14
ICS B24B057-02; C08K003-00; C08L033-02; C08L071-02; C09K003-14; H01L021-304

CC 76-3 (Electric Phenomena)

09/16/2005 GOUDREAU 10/807139 Page 55 Section cross-reference(s): 38 FAN.CNT 1 APPLICATION NO. PATENT NO. KIND DATE DATE -----\_ \_ \_ \_ \_ \_ -----\_\_\_\_ -----PΤ JP 2000313815 A2 20001114 JP 1999-124498 19990430 PRAI JP 1999-124498 19990430 CLASS CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. ---------JP 2000313815 ICM C08L101-14 ICS B24B057-02; C08K003-00; C08L033-02; C08L071-02; C09K003-14; H01L021-304 AB The agents contain water-soluble polymers which show ≥50% adsorption (at 25°) on the abrasive particles when 0.05 weight part of the polymers are added to 100 weight parts aqueous slurry containing 10 weight% abrasive particles having primary particle size 10-100 nm. Preferably, the polymers are polyoxyalkylenes or vinyl copolymers. High accuracy and good surface appearance are achieved by high-speed polishing of semiconductor substrates or devices with the compns. containing the polymers above and abrasives. ST semiconductor chem mech polishing polyoxyalkylene; water soluble vinyl polymer abrasive semiconductor IT Abrasives Semiconductor device fabrication (abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) Polyoxyalkylenes, properties IT RL: PRP (Properties); TEM (Technical or engineered material use); USES (abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) Polyoxyalkylenes, properties IT RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic; abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) IT Polishing (chemical-mech.; abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) IT 305383-98-6P 305383-99-7P 305384-01-4P 305384-02-5P 305384-04-7P 305384-06-9P 305384-08-1P 305807-95-8P, Ethylene oxide-propylene oxide block copolymer monoacrylate-methacrylic acid graft copolymer ammonium salt 312314-72-0P 312314-75-3P 312746-76-2P 312746-77-3P 312746-82-0P RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor

substrates or devices)

IT 25322-68-3, Polyethylene glycol 106392-12-5, Ethylene oxide-propylene oxide block copolymer RL: PRP (Properties); TEM (Technical or engineered material use); USES

(Uses) (abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor

substrates or devices) IT 7631-86-9, Silica, uses 11129-18-3, Cerium oxide RL: TEM (Technical or engineered material use); USES (Uses) (abrasive; abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) 106392-12-5, Ethylene oxide-propylene IT oxide block copolymer RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses) (abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) RN 106392-12-5 HCAPLUS CNOxirane, methyl-, polymer with oxirane, block (9CI) (CA INDEX NAME) CM CRN 75-56-9 CMF C3 H6 O CM 2 CRN 75-21-8 CMF C2 H4 O IT 7631-86-9, Silica, uses RL: TEM (Technical or engineered material use); USES (Uses) (abrasive; abrasive compns. containing water-soluble polymers for chemical-mech. polishing of semiconductor substrates or devices) 7631-86-9 HCAPLUS RN (CA INDEX NAME) Silica (6CI, 7CI, 8CI, 9CI) CN o = si = oL39 ANSWER 30 OF 34 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN 2000-587095 [55] AN WPIX N2000-434529 DNN DNC C2000-174953 ΤI Chemical mechanical polishing of low dielectric constant polymer surface of integrated circuit wafer, utilizes slurry comprising fine metal oxide particles uniformly dispersed in stable aqueous medium. DC A85 L03 U11 IN HOSALI, S D; SACHAN, V

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(HOSA-I) HOSALI S D; (SACH-I) SACHAN V; (RODE-N) RODEL HOLDINGS INC
PA
CYC
     24
                     A1 20000824 (200055)* EN
PΙ
     WO 2000049647
                                                18
        RW: AT BE CH CY DE DK ES FI FR GB GR IE IT LU MC NL PT SE
         W: CN JP KR SG
     US 2001013507
                     A1 20010816 (200149)
                                                       C23F001-00
     EP 1171906
                     A1 20020116 (200207)
                                          EN
                                                      H01L021-302
         R: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE
     KR 2001111261
                     A 20011217 (200238)
                                                      H01L021-304
     JP 2002537652
                     W 20021105 (200304)
                                                16
                                                      H01L021-304
     WO 2000049647 A1 WO 2000-US3893 20000216; US 2001013507 A1 Provisional US
     1999-120567P 19990218, CIP of US 2000-505042 20000216, US 2000-742853
     20001221; EP 1171906 A1 EP 2000-913478 20000216, WO 2000-US3893 20000216;
     KR 2001111261 A KR 2001-710397 20010816; JP 2002537652 W JP 2000-600297
     20000216, WO 2000-US3893 20000216
    EP 1171906 Al Based on WO 2000049647; JP 2002537652 W Based on WO
     2000049647
                          19990218; US 2000-505042
PRAI US 1999-120567P
                                                         20000216;
     US 2000-742853
                          20001221
     ICM C23F001-00; H01L021-302; H01L021-304
     ICS B24B037-00; C09K003-14; H01B013-00
     WO 200049647 A UPAB: 20001102
AB
     NOVELTY - A low dielectric constant polymer surface of an integrated
     circuit (IC) wafer is chemically and mechanically
     polished with a chemical mechanical
     polishing slurry comprising a colloidally stable
     dispersion of metal oxide particles. The particles have medium having
     40-430 m2/g surface area, less than 1 mu m aggregate size distribution,
     and less than 0.4 mu m mean aggregate diameter.
          USE - The method is useful in chemical mechanical
     polishing a low dielectric constant polymer surface of an
     integrated circuit wafer or semiconductor devices. It is useful in
     chemical mechanical planarization to remove uneven inter-level dielectric
     (ILD) topography, layers of material, surface defects including scratches,
     roughness, or contaminant particles, e.q. dirt or dust.
          ADVANTAGE - The utilization of the slurry provides an
     effective polishing while minimizing surface imperfections and
     defects.
     Dwg.0/0
     CPI EPI
FS
FA
    AB
MC
     CPI: A09-A03; A11-C04; A12-E07C; L04-B04; L04-C07; L04-C12E
     EPI: U11-C05B9A; U11-C05D1; U11-C06A1A
L39
    ANSWER 31 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN DUPLICATE 2
AN
     2000:147946 HCAPLUS
DN
     132:230589
ED
     Entered STN: 05 Mar 2000
ΤI
     Surfactant based alumina slurries for copper CMP
ΑU
    Babel, Ashok K.; Mackay, Raymond A.
CS
    Center for Advanced Materials Processing, Clarkson University, Potsdam,
    NY, 13699, USA
SO
    Materials Research Society Symposium Proceedings (2000),
    566 (Chemical-Mechanical Polishing--Fundamentals and Challenges), 135-142
    CODEN: MRSPDH; ISSN: 0272-9172
PB
    Materials Research Society
\mathtt{DT}
    Journal
LΑ
    English
CC
     76-14 (Electric Phenomena)
     Section cross-reference(s): 56, 57
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AB The polishing of copper and examination of the polished surfaces were carried out with surfactant based alumina slurries to yield interesting results. Contrary to our expectation and previously reported research, some of the surfactant based alumina slurries resulted in higher copper polish rates when compared to the control. Of the nonionic surfactants, BrijR 35 was overall the most effective in both acidic and basic media. effective at the pH for the appropriate charge type. For the range of surfactants studied, polish rates correlated with the HLB of the nonionic surfactants. The Hydrophile-Lipophile Balance (HLB) is related to the solubility of the surfactant, with higher number corresponding to increased water dispersibility. The surfactant BrijR 35, with the nonionic composition polyoxyethylene(23) lauryl ether, resulted in a dramatic improvement in the average surface uniformity when compared with the control at pH 2, and Sodium Dodecyl Sulfate produced even greater uniformity. Addnl., the effect of BrijR 35 surfactant was maintained with change in abrasive size, pad and polishing tool. In order to insure that surfactants are compatible with the chemical reagents contained in the com. slurries , two chemistries (ferric nitrate and hydrogen peroxide) were employed to test the efficiency of the selected surfactants in their presence. The results showed that the effect of surfactant on stability and removal rate is not influenced by the presence of the chems. Preliminary results indicate that surfactants can have a beneficial effect on both defects and post polish clean.

ST aluminum slurry surfactant abrasive copper CMP

IT

(alumina; surfactant based alumina slurries for copper CMP)

ITSurfactants

> (in abrasive slurry; surfactant based alumina slurries for copper CMP)

IT Surfactants

(nonionic; surfactant based alumina slurries for copper CMP)

ΙT Abrasives

(slurry; surfactant based alumina slurries for copper CMP)

Hydrophile-lipophile balance value IT

(surfactant based alumina slurries for copper CMP)

IT 1344-28-1, Alumina, properties

> RL: NUU (Other use, unclassified); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(slurry; surfactant based alumina slurries for copper CMP)

IT 7440-50-8, Copper, properties 7722-84-1, Hydrogen peroxide, properties 10421-48-4, Ferric nitrate

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(surfactant based alumina slurries for copper CMP)

RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Adler, J; Mat Res Soc Symp Proc 1998, V501, P387 HCAPLUS
- (2) Anon; Private Communication from BJ Palla, D O Shah, M Bielmann, and RK Singh
- (3) Anon; Private Communication from G Sabde
- (4) Babel, A; submitted for publication
- (5) Bielmann, M; Electrochemical and Solid-state Letters 1999, V2(3), P148 **HCAPLUS**
- (6) Braun, A; Semiconductor International 1998, V21, P65

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(7) Campbell, D; CAMP Newsletter 1994, V10, P1
(8) Free, M; Micro 1998, May, P29
L39
    ANSWER 32 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
     1998:242076 HCAPLUS
DN
     128:285577
ED
    Entered STN: 29 Apr 1998
    Cerium oxide abrasive and polishing of substrates
TI
IN
    Yoshida, Masato; Matsuzawa, Kiyoshi
PA
    Hitachi Chemical Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 4 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
     ICM C09K003-14
     ICS C01F017-00; C08K003-22; C08L101-00; C09C001-68; H01L021-304
CC
     57-7 (Ceramics)
     Section cross-reference(s): 76
FAN.CNT 1
     PATENT NO.
                        KIND
                             DATE
                                         APPLICATION NO.
                                                                 DATE
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                                           -----
    JP 10102040
                        A2
                               19980421
                                           JP 1996-258769
                                                                 19960930
PRAI JP 1996-258769
                               19960930
CLASS
 PATENT NO.
                CLASS PATENT FAMILY CLASSIFICATION CODES
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JP 10102040
                ICM
                       C09K003-14
                       C01F017-00; C08K003-22; C08L101-00; C09C001-68;
                ICS
                       H01L021-304
AB
    Ce oxide particles containing ≥90% primary particles showing contours
    containing angular parts smaller than 120° in observation by
    transmission type electron microscope are dispersed in water, and the Ce
    oxide particle slurry is used as abrasive for
    polishing of substrates optionally having SiO2 insulating layer.
    Optionally, the slurry contains a dispersant selected from
    water-soluble organic polymer, water-soluble anionic surfactant,
    water-soluble nonionic surfactant, and water-soluble amine.
    cerium oxide abrasive silicon substrate polishing
ST
IT
    Surfactants
        (anionic, dispersant; polishing of silicon substrate by
       slurry-form cerium oxide abrasive containing)
IT
    Amines, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
       (dispersant; polishing of silicon substrate by slurry
       -form cerium oxide abrasive containing)
IT
    Surfactants
        (nonionic, dispersant; polishing of silicon
       substrate by slurry-form cerium oxide abrasive
       containing)
    Polishing
TT
       (of silicon substrate; slurry-form cerium oxide
       abrasive for)
IT
    Abrasives
       (slurry-form cerium oxide abrasive for
       polishing silicon substrate)
IT
    9003-03-6, Ammonium polyacrylate
    RL: TEM (Technical or engineered material use); USES (Uses)
       (dispersant; polishing of silicon substrate by slurry
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-form cerium oxide abrasive containing)

7631-86-9, Silica, uses

IT

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RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing of silicon substrate coated with; slurry
        -form cerium oxide abrasive for)
IT
     7440-21-3, Silicon, processes
     RL: PEP (Physical, engineering or chemical process); PROC (Process)
        (polishing of substrates of; slurry-form cerium
        oxide abrasive for)
     1306-38-3, Cerium oxide, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (slurry-form cerium oxide abrasive for
        polishing silicon substrate)
L39
     ANSWER 33 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
     1997:248770 HCAPLUS
DN
     126:350637
ED
     Entered STN: 17 Apr 1997
ΤI
     A study of post-chemical-mechanical-polish cleaning strategies
ΑU
     Huynh, C.; Rutten, M.; Cheek, R.; Linde, H.
CS
     Microelectronics Div., IBM, Essex Junction, VT, 05452, USA
SO
     Proceedings - Electrochemical Society (1997), 96-22 (Chemical Mechanical
     Planarization), 16-26
     CODEN: PESODO; ISSN: 0161-6374
PB
     Electrochemical Society
DT
     Journal
LA
     English
CC
     77-3 (Magnetic Phenomena)
     Section cross-reference(s): 66
AB
     Chemical Mech. Polishing (CMP) has emerged as the premier
     technique for achieving both local and global planarization. One of the
     primary concerns in the use of CMP, however, is the efficient
     and complete removal of CMP contaminants such as slurry
     and residual hydrocarbons. This paper discusses the removal of
     silica-based slurries utilized for polysilicon and oxide
     CMP processes. The effects of mech. brush cleaning, chemical
     treatments, and polish processes on defect d. for a 16Mb memory
     technol. are presented. In addition, the chemical compatibility of
     polishing slurries with various brush and
     polishing pad materials is discussed.
ST
     silicon wafer chem mech polishing cleaning
IT
     Memory devices
        (RAM (random access); post-chemical-mech.-polish cleaning
        strategies for silicon wafers)
ΙT
     Polishing
        (chemical-mech.; post-chemical-mech.-polish cleaning strategies for
        silicon wafers)
IT
     Surfactants
        (nonionic; post-chemical-mech.-polish cleaning
        strategies for silicon wafers)
     Cleaning
IT
     Semiconductor devices
        (post-chemical-mech.-polish cleaning strategies for silicon
        wafers)
     Abrasives
IT
        (slurry removal; post-chemical-mech.-polish cleaning
        strategies for silicon wafers)
     7440-21-3, Silicon, processes
IT
     RL: DEV (Device component use); PEP (Physical, engineering or chemical
     process); PROC (Process); USES (Uses)
        (post-chemical-mech.-polish cleaning strategies for silicon
        wafers)
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IT 75-59-2, TMAH

RL: PEP (Physical, engineering or chemical process); PROC (Process) (post-chemical-mech.-polish cleaning strategies for silicon wafers)

IT 7631-86-9, Silica, processes

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)

(slurry removal; post-chemical-mech.-polish cleaning strategies for silicon wafers)

RE.CNT 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Ali; Microcontamination 1994
- (2) Ali, I; Semiconductor Intl 1990, P92 HCAPLUS
- (3) Blackwell, R; US 5320706
- (4) Bowling, R; J Electrochem Soc 1995, V137, P2208
- (5) Cook, L; J Non-Crystalline Solids 1990, V120, P152 HCAPLUS
- (6) Huynh; "CMP Clean Process Engineering, CMP Clean Evaluation," Presented at Sematech Surface Prep PTAB Meeting 1993
- (7) Huynh, C; 1993 Research/TP Symposium on Silicon Technology 1993
- (8) Huynh, C; U S Patent Pending
- (9) Iler, R; The chemistry of Silica 1979, P366
- (10) Jeon, J; Belgium Semiconductor Symposium 1995
- (11) Park, J; Microcontamination Conference 1992
- (12) Roy; J Electrochem Soc 1995, V142, P216 HCAPLUS
- (13) Singer, P; Wafer Processing News, Semiconductor Intl 1995
- IT 75-59-2, TMAH

RL: PEP (Physical, engineering or chemical process); PROC (Process) (post-chemical-mech.-polish cleaning strategies for silicon wafers)

- RN 75-59-2 HCAPLUS
- CN Methanaminium, N,N,N-trimethyl-, hydroxide (9CI) (CA INDEX NAME)

OH-

- L39 ANSWER 34 OF 34 HCAPLUS COPYRIGHT 2005 ACS on STN
- AN 1995:582574 HCAPLUS
- DN 122:320762
- ED Entered STN: 02 Jun 1995
- TI Synthetic diamond-based polishing suspension for semiconductors
- IN Komarov, Vitaly Fedorovich; Sakovich, Gennady Viktorovich; Petrov, Evgeny Anatolievich; Klimov, Anatoly Valentinovich; Kostjukov, Sergei Ivanovich; Baraboshkin, Konstantin Sergeev
- PA Nauchno-Proizvodstevennoe Obiedinenie "Altai", Russia
- SO PCT Int. Appl., 18 pp. CODEN: PIXXD2
- DT Patent
- LA Russian
- IC G09G001-02; C09G001-08

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CC
    57-6 (Ceramics)
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                        APPLICATION NO.
                                                                DATE
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                                         -----
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    WO 9422970
ΡI
                        A1
                              19941013 WO 1994-RU68
                                                                19940401
        W: BY, CA, JP, UA, US
        RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
     RU 2034889
                      C1 19950510
                                        RU 1993-12940
PRAI RU 1993-12940
                        Α
                              19930402
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _____
               ____
WO 9422970 IC G09G001-02IC WO 9422970 ECLA C09K003/14B
                      G09G001-02IC C09G001-08
    The invention concerns a polishing compound for giving a
    superfinish to surfaces, containing 5-10% of an abrasive component suspended
    and a liquid medium. The abrasive component is a synthetic diamond-containing
    material with specified properties in which the primary particles are 4-6
    nm in size and combine to form aggregates of 20-500 nm in size, with sp.
    surface areas of 250-450 m2/g and pore volume of 0.6-1.0 cm3/g. A typical
    composition contained the above abrasive material 5-10, glycerol or
    diethylene glycol 10-15, H2O2 5-15, ethylenediamine 0.1-1.0, and KOH or
    NaOH 1-3% in H2O.
ST
    diamond synthetic polishing suspension; glycerol synthetic
    diamond polishing suspension; hydrogen peroxide diamond
    polishing suspension; hydroxide synthetic diamond
    polishing suspension
IT
    Polishing materials
        (synthetic diamond-based polishing suspension for
       semiconductors)
IT
    Petrolatum
    Waxes and Waxy substances
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
        (synthetic diamond-based polishing suspension for
       semiconductors)
IT
    Polishing
        (chemical-mech., synthetic diamond-based
       polishing suspension for semiconductors)
    Alcohols, uses
IT
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
        (long-chain, ethoxylated, synthetic diamond-based polishing
       suspension for semiconductors)
IT
    148-24-3, 8-Hydroxyquinoline, uses
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
       (saturated aqueous solution; synthetic diamond-based polishing
       suspension for semiconductors)
    56-81-5, Glycerol, uses 107-15-3, Ethylenediamine, uses 111-46-6,
IT
    Diethylene glycol, uses 112-80-1, Oleic acid, uses 1310-58-3,
    Potassium hydroxide, uses 1310-73-2, Sodium hydroxide, uses
    7631-86-9, Aerosil, uses 7722-84-1, Hydrogen peroxide, uses
    9003-11-6, Ethylene oxide-Propylene
    oxide copolymer 11099-07-3, Stearin 12751-48-3, Syntanol
    25322-68-3D, Polyethylene glycol, ethers, with fatty alc.
    Glycerol monostearate
    RL: MOA (Modifier or additive use); TEM (Technical or engineered material
    use); USES (Uses)
       (synthetic diamond-based polishing suspension for
```

semiconductors)

IT 7782-40-3, Diamond, uses

RL: TEM (Technical or engineered material use); USES (Uses) (synthetic; synthetic diamond-based **polishing** suspension for semiconductors)

IT **7631-86-9**, Aerosil, uses

RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(synthetic diamond-based polishing suspension for

semiconductors)

RN 7631-86-9 HCAPLUS

CN Silica (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

o = si = o

=> SAVE L39 GOU807/A ANSWER-SET L39 HAS BEEN SAVED AS 'GOU807/A'